



**THE STRUCTURE OF MIXED β -ARYL(FURYL)BENZOIN,
2-HYDROXY-2-(4''-CHLOROPHENYL)-1-(5'-*N,N*-
DIMETHYLHYDRAZONYLFURYL-2')ETHANONE-1**

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The structure of the thermal $\alpha \rightarrow \beta$ benzoin isomerization product, 2-hydroxy-2-(4''-chlorophenyl)-1-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 has been proved by the XRD study. The possibility of the Me₂NN=CH-substituent conjugation with the carbonyl group via the furane ring take place due to the overall planarity this molecule fragment.

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The α -benzoin **1** and β -benzoin **2** structures were established by the data of NMR ¹H and MS spectra^[1]. But the β -benzoin **2** structure remained unstudied.

EXPERIMENTAL

2-Hydroxy-2-(4''-chlorophenyl)-1-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 (**2**)

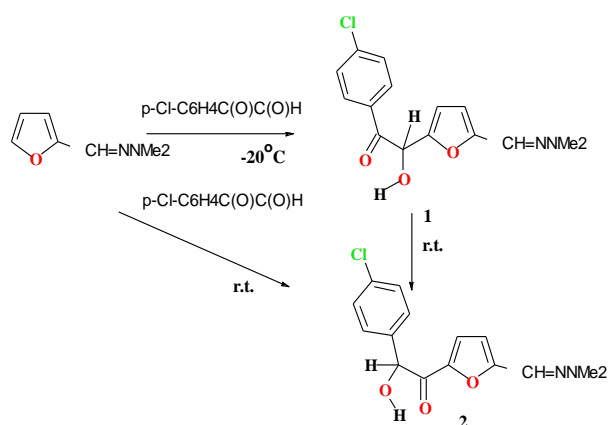
The solution of *N,N*-dimethylhydrazone of 2-furanecarbaldehyde (13.0 mmol, 1.795 g) in PhH (5 ml) was added to the boiling solution of 4-chlorophenylglyoxal hydrate (10.0 mmol, 1.866 g) in PhH (20 ml). The reaction mixture was boiled during 30 min, than it was kept at 20 °C during 2 days, than PhH was evaporated *in vacuo*, the residue was washed hexane (20 ml), PhH (15 ml), yielding 2.168 g (70 %) 2-hydroxy-2-(4''-chlorophenyl)-1-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 **2**, orange crystals, mp. 145-146 °C, after crystallization from CH₂Cl₂ mp. 150-151 °C (cf. with mp. 150-151 °C¹), identify with the sample of **2**^[1] by NMR ¹H and MS.

The crystals of **1** were grew from CH₂Cl₂, monoclinic, C₁₅H₁₇N₂O₃Cl·0.25(CH₂Cl₂), at 298 K, $a = 26.4863(19)$ Å, $b = 5.7593(5)$ Å, $c = 11.2223(9)$ Å, $\beta = 103.999(8)^\circ$, $V = 1661.0(2)$ Å³, $M_r = 325.01$, $Z = 4$, space group C2, $d_{\text{calc}} = 1.300$ g/cm³, $\mu(\text{MoK}\alpha) = 0.245$ mm⁻¹, $F(000) = 680$. Data were measured using Xcalibur 3 diffractometer ($T=298$ K, graphite-monochromated MoK α radiation, $2\theta/\theta$ scan, $2\theta_{\text{max}}=58.36^\circ$).

The structures were solved by direct method using the SHELXTL PLUS program package.³ Refinement against F^2 in an anisotropic approximation (the hydrogen atoms isotropic in the riding model with $U_{\text{iso}}=nU_{\text{eq}}$ of worn atom, $n=1.5$ for HO-group and Me-group, $n = 1.2$ for other hydrogen atoms) by a full matrix least-squares method for 6445 reflections was carried out to $wR_2=0.136$ ($R_1=0.071$ for 1887 reflections with $F > 4\sigma(F)$, $S=1.07$).

INTRODUCTION

Earlier we had reported that 4-chlorophenylglyoxal reacted with *N,N*-dimethylhydrazone of 2-furanecarbaldehyde yielding at -20 °C in ether solution the α -benzoin, 2-hydroxy-1-(4''-chlorophenyl)-2-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 **1**, which spontaneously isomerized at room temperature in β -benzoin, 2-hydroxy-2-(4''-chlorophenyl)-1-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 **2**.¹ The β -benzoin **2** is only product of the 4-chlorophenylglyoxal reaction with *N,N*-dimethylhydrazone of 2-furanecarbaldehyde at room temperature.¹ The letter " α " had been used to indicate the less stable isomer, the letter " β " had been used to indicate the more stable isomer.²



Scheme 1

RESULTS AND DISCUSSION

The XRD study data are evidence of β -benzoin structure of 2-hydroxy-2-(4''-chlorophenyl)-1-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 **2** (Figure 1, Tables 1, 2). The Me₂NN=CH moiety conjugates with the benzoin carbonyl group via the furan ring bonds. The Me₂NN=CH moiety, the furan ring and the carbonyl group are oriented in the same plane, whereby this conjugation becomes possible. The middle quadratic deviation of these atoms from the plane of the conjugated bonds is equal 0.036 Å.

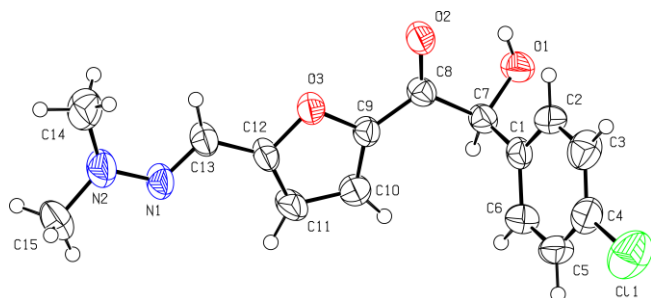
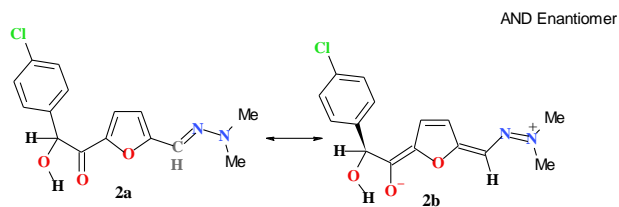


Figure 1. The structure of 2-hydroxy-2-(4''-chlorophenyl)-1-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 **2**.

This conjugation chain arises to charge transfer from nitrogen N(2) atom on O(2) oxygen atom. The structure of 2-hydroxy-2-(4''-chlorophenyl)-1-(5'-*N,N*-dimethylhydrazonofuryl-2')-ethanone-1 **2** can be more effectively described by the structure **2b** than the structure **2a** (Scheme 2).



Scheme 2

It was found that in β -benzoin **2** N(2) atom has the nearly planar configuration. The sum of bond angles centered at this nitrogen atom ($\Sigma\beta$) is 359°. The N(1)-N(2) bond is shortened to 1.322(4) Å (cf. with N-N bond length 1.45 Å⁴). The O(2)-C(8) bond is some elongated to 1.236 (5) Å (cf. with for C=O bond length 1.21 Å⁴).

The C(8)-C(9) bond and the C(12)-C(13) bond are some shortened to 1.433(5) Å and 1.431(5) Å relatively toward ordinary C(sp²)-C(sp²) bond (1.47 Å.⁴¹ All these structure data mean the domination part of the resonance form **2b** in the β -benzoin **2** structure. It may suppose that arising of the long chain of the conjugation between Me₂N- and C=O-moiety is the moving force of the thermal $\alpha \rightarrow \beta$ benzoin isomerization yielding β -benzoin **2**.¹

The *para*-chlorophenyl substituent is perpendicular oriented to this conjugation plane (the C1-C7-C8-C9 torsion angle is -89.7(4)°).

In the resolved crystal of β -benzoin **2** the C(7) atom has absolute *S* configuration.

In the crystals molecules of β -benzoin **2** are linked in the chains along the *b* crystallographic direction due to intermolecular bifunctional hydrogen bonds O(1)-H(1)...O(1ⁱ) [i: 3/2-x, -1/2+y, 2-z] (H...O 2.33 Å, O-H...O 133°) and O(1)-H(1)...O(2ⁱ) (H...O 2.14 Å, O-H...O 150°). Also in the crystal σ -hole bond Cl(1)...O(3ⁱⁱ) [ii: 3/2-x, -1/2+y, 1-z] (Cl...O 3.18, C(4)-Cl(1)...O(3) 167°, C(12)-O(3)...Cl(1) 101°) take place.

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