



## Application of Box-Behnken Design for Formulation and Optimization of Novel Gastroretentive Microballoons by Solvent Evaporation Method.

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### ABSTRACT:

The present study involves preparation and evaluation of floating microballoons of Repaglinide for improving the bioavailability. Microballoons promises to be a potential approach for gastric retention time in the gastrointestinal tract (GIT) they are designed to overcome some of the problems of conventional therapy and enhance the therapeutic efficacy of a given drug by controlling release rate for drugs. The Microballoons were prepared by Solvent Evaporation technique using polymers such as Ethyl Cellulose, Hydroxy Propyl Methyl Cellulose along with solvents like alcohol, dichloromethane and Surfactant like Tween 80 to decrease the interfacial tension between fluids. The optimization of microballoons was done by using Design of Experiments (DoE) studies Box-Behnken design is utilised for the same. Prepared microballoons were analyzed for their Percentage Buoyancy (PB), Mean Geometric Diameter (DG) and Entrapment Efficiency (EE). *In vitro* drug release studies were performed and drug release kinetics was evaluated using the linear regression method. *In vitro* drug release of the optimised formulation was found to be 95.48 %. The prepared microballoons exhibited prolonged drug release i.e. Upto 12 hours. Stability studies was carried out at  $40 \pm 2^\circ\text{C}$  Temp and  $75 \pm 5\%$  Relative Humidity shows no significant change in microballoons of the optimized formulation after 06 months of storage.

**Keywords:** Floating Microballoons, Hydroxy Propyl Methyl Cellulose, Ethyl Cellulose, Repaglinide, Percentage Buoyancy (PB), Mean Geometric Diameter (DG) and Entrapment Efficiency (EE).

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## INTRODUCTION:

Oral dosage forms face several restrictions like inability to retain the dose in GIT (Gastro Intestinal Track) due to fluctuation in gastric emptying which will lead to non-uniform absorption, inadequate medication & shorter residence time of dosage form in stomach these complications provoked to the development of control release dosage form with Gastroretentive properties which lead to formulation of Microballoons with low density, sufficient buoyancy to float over gastric contents and remain in stomach for prolonged period of time.<sup>[1,2]</sup> As the system floats over gastric contents the drug is released slowly at desired rate resulting in increased gastric retention with reduced fluctuations in plasma drug concentration. Microballoons hold potential approach for gastric retention due to significant increase in gastric residence time, enhancement in bioavailability of BCS Class II Drugs e.g., Repaglinide, Glibenclamide, Glibornuride, Gliclazide, Glipizide, Gliquidone, Glisoxepide and Glycopyramide, improving patient compliance by reducing dosing frequency, enhance retention of medication which solubilize only in stomach, enhance solubility for drugs which are less soluble at higher pH.

Microballoons (Hollow microsphere) are drug delivery system that promises to be a potential approach for gastric retention. Microballoons are based on non-effervescent system containing empty particles of spherical shape without core. Microballoons drug delivery systems have shown to be of better significance in controlling release rate for drugs having site specific absorption. The floating microballoons showed gastro retentive controlled release delivery with efficient means of enhancing the bioavailability and promises to be a potential approach for gastric retention. Optimized hollow Microballoons will find the central place in novel drug delivery, particularly in safe, targeted and effective *in-vivo* delivery promises to be a potential approach for gastric retention.<sup>[3, 4]</sup> Floating microballoons can be potential carrier to achieve sustained delivery of drug in stomach by increasing residence time of drug and slowly releasing drug at its site of absorption. The drugs which are specifically absorbed from upper part of GIT or which are unstable at alkaline lower GIT can be formulated as floating microballoons to increase their absorption and thus bioavailability. Delivering drugs through microballoons can help maintain plasma concentration of drug without being fluctuated for longer time. In present study, Microballoons of Repaglinide were prepared, optimized and evaluated for in vitro performance and stability study.

## MATERIALS AND METHODS:

Repaglinide was procured from Chempur Pharma, Mumbai, Hydroxy PropylMethyl Cellulose K4M and Hydroxy Propyl Methyl Cellulose K15M, Ethyl Cellulose, Tween 80 was purchased from Ranbaxy Fine chemicals, Mumbai. All other chemicals used were of analytical grade.

### **Preparation of Microballoons**

Microballoons were prepared by solvent evaporation technique.<sup>[3]</sup> Repaglinide (50 mg) and HPMCK15M and EC were used in different ratios. The total weights of HPMC and EC per 100 mg of Repaglinide were kept constant at 200 mg. The polymers were dissolved in a mixture of alcohol and dichloromethane (in varying ratios). The total amount of solvent mixture was kept constant at 100 ml and the procedure was carried out at room temperature. The resulting solution was poured into 250 ml of distilled water containing 0.01 % v/v Tween 80, maintained at different temperatures, and then stirred at varying agitation speed (200 - 1600 rpm) for 20 min to allow the volatile solvent to evaporate. The microballoons formed were filtered, washed with distilled water and dried.

### **Selection of Polymers**

Floating microballoons prepared using polymers as Sodium Alginate; Methocels (HPMC) and Ethyl cellulose are being researched for controlled delivery of drug in stomach. For optimization of type of polymers and their concentrations, various trials were performed. Other parameters like stirring speed (500 rpm), temperature (28°C) and phase ratio were kept constant during these selection studies. The selection criteria were; Percentage Buoyancy (PB), Mean Geometric Diameter (DG) and Entrapment Efficiency (EE).

### **Selection of processing conditions**

#### **1. Selecting of stirring speed**

Optimized batches of microballoons with selected excipients were subjected to various stirring speeds (200-1600 rpm) at room temperature. All the batches of prepared microballoons were analyzed for their PB, DG and EE.

#### **2. Selection of temperature**

To study the effect of temperature, the optimized batches of microballoons were prepared at various temperature conditions. For all these batches, excipients and stirring speed were kept constant and evaluated for three constraints as PB, DG and EE.

#### **3. Selection of phase ratio**

Different ratios of good solvent and poor solvent were utilized and the resulting microballoons were evaluated for various parameters like; PB, DG and EE. For all these batches, excipients, stirring speed and temperature were kept constant.

### **Optimization by Experimental Design**

The optimizations of microballoons were further persisted by using Design of Experiments (DoE).<sup>[4]</sup> On the basis of preliminary trials, experimental domains were determined and appropriate experimental design was selected. Box-Behnken design was implemented for Repaglinide microballoons.<sup>[5]</sup>

#### **Box-Behnken design for Repaglinide microballoons:-**

Based on preliminary trials, independent variables (factors) were determined as; Amount of EC ( $X_1$ ), Amount of Methocel K15M ( $X_2$ ) and stirring speed ( $X_3$ ) (Table 1).

**Table 1: Design layout of Box-Behnken design batches**

Batch Code	Independent Variables		
	Amount of EC (X <sub>1</sub> )	Amount of MethocelK15M (X <sub>2</sub> )	Stirring speed (X <sub>3</sub> )
B1	-1	-1	0
B2	1	-1	0
B3	-1	1	0
B4	1	1	0
B5	-1	0	-1
B6	1	0	-1
B7	-1	0	1
B8	1	0	1
B9	0	-1	-1
B10	0	1	-1
B11	0	-1	1
B12	0	1	1
B13	0	0	0

In this design, three factors with three levels were probed to investigate the main and interaction effects on preferred responses. The design consisted a total 13 runs (B1 to B13) and each of them was formulated in triplicates in order to estimate reproducibility of the model. [5] A second order quadratic model incorporating interactive and polynomial terms was exercised to evaluate the responses.

$$Y_1 = b_0 + b_1X_1 + b_2X_2 + b_3X_3 + b_{12}X_1X_2 + b_{23}X_2X_3 + b_{13}X_1X_3 + b_{11}X_1^2 + b_{22}X_2^2 + b_{33}X_3^2 \quad (1)$$

Where,

Y<sub>1</sub> was dependent variable,

b<sub>0</sub> was arithmetic mean response of 13 runs and

b<sub>1</sub> was the estimated coefficient for factor X<sub>1</sub>.

The main effects (X<sub>1</sub>, X<sub>2</sub> and X<sub>3</sub>) signify average result of altering one factor at a time from its lowest to highest value. The interaction terms (X<sub>1</sub>X<sub>2</sub>, X<sub>2</sub>X<sub>3</sub> and X<sub>1</sub>X<sub>3</sub>) prompt change in responses when two factors are simultaneously altered. The polynomial terms (X<sub>1</sub><sup>2</sup>, X<sub>2</sub><sup>2</sup> and X<sub>3</sub><sup>2</sup>) were added to investigate nonlinearity of the model. [6]

Data were further analyzed by Microsoft Excel version 2019 for regression analysis. Analysis of variance (ANOVA) was executed to assure no significant difference between developed full model and reduced model. Contour, response surface and perturbation plots were generated to study response variations against independent variables using Design Expert<sup>®</sup> 13 (Stat-Ease, Inc. Minneapolis, USA) software. Additionally the composition of optimized (check point) batch was derived by constructing overlay plot. The percentage relative error of each response was calculated using following equation in order to judge validity of the model. [7]

### Evaluation of Microballoons

#### 1. Drug Content, Percent Yield and Entrapment Efficiency:

Accurately weighed quantity of microballoons was dissolved in sufficient quantity of a suitable solvent, in which it was easily soluble, in a volumetric flask. The volume was

made up to the mark with the same solvent. These solutions were appropriately diluted and drug content was determined by respective analytical methods. The experimental drug content was calculated using calibration equation as a mean of three independent determinations. The percent (%) yield and Entrapment Efficiency of sample was calculated using Equation 2 and 3.<sup>[8, 9]</sup> Average of three determinations was considered as result of the test carried out.

$$\% \text{ Yield} = \frac{\text{total weight of microballons}}{\text{total weight of drug and excipients}} \times 100 \quad (2)$$

$$\text{Entrapment efficiency} = \frac{\text{Weight of repaglinide in microballons}}{\text{Initial weight of Repaglinide}} \times 100 \quad (3)$$

## 2. Size Analysis

Size analysis of prepared microballoons was performed using optical microscopy method. The size of randomly selected microballoons (300) was measured and their mean geometric diameter (DG) was calculated.<sup>[10]</sup>

## 3. Dissolution method

In-vitro release study of prepared microballoons was carried out in USP type II (paddle type). The dissolution study was carried out in 900 mL dissolution media at  $37 \pm 0.5^\circ\text{C}$ . Sample of 5 mL were withdrawn at specified time interval and replaced with fresh media. The samples were analyzed using suitable analytical method in order to find out amount of dissolved drug. All determinations were performed in triplicate.<sup>[11, 12]</sup>

## 4. Stability Study

According to the ICH guidelines, the optimized batches were placed in high density polyethylene (HDPE) bottles. The mouths of the bottles were closed tightly with aluminium foil to prevent the access of air from the atmosphere to the sample inside the bottles. Samples were stored at  $40 \pm 0.5^\circ\text{C}$  and  $75 \pm 5\%$  relative humidity (RH) for 6 months in programmable environmental test chamber. The samples were withdrawn at particular time period and statistically analyzed.<sup>[13]</sup>

## Results and Discussion

### Selection of excipients and processing conditions

#### 1. Selection of Excipients

In the formulation of pharmaceutical dosage form various excipients are utilized to improve the functionality of dosage forms. All these excipients have a significant impact on their bioavailability, toxicity, stability and efficacy.<sup>[14, 15]</sup> Addition of small amounts of surfactants or polymers is the method usually applied to obtain the best spherical microballoons. Surfactants or polymers can decrease the interfacial tension between fluids, change the viscosity of phases and hence influences the droplets sphericity. The results of Repaglinide microballoons with all studied excipients have been summarized in

Table 2. Concentrations of excipients incorporated in formation of microballoons were within limits as per inactive ingredients guidelines (IIG) of FDA. <sup>[16]</sup>

**Table 2: Selection of excipients for repaglinidemic microballoons**

Batch	Excipients	Amount (mg)	Results *		
			PB (%)	DG ( $\mu$ m)	EE (%)
PR1	Sodium alginate	50	36.15 $\pm$ 2.64	247.33 $\pm$ 17.52	25.41 $\pm$ 2.12
PR2		100	33.45 $\pm$ 2.58	328.57 $\pm$ 25.24	28.34 $\pm$ 1.78
PR3		150	40.12 $\pm$ 3.45	385.56 $\pm$ 35.36	28.95 $\pm$ 3.29
PR4		200	42.57 $\pm$ 3.75	378.54 $\pm$ 37.51	35.93 $\pm$ 2.57
PR5		250	43.45 $\pm$ 5.56	453.87 $\pm$ 54.21	43.87 $\pm$ 2.42
PR6	Ethyl cellulose	50	40.52 $\pm$ 2.04	283.54 $\pm$ 24.56	29.27 $\pm$ 1.93
PR7		100	49.63 $\pm$ 3.52	278.71 $\pm$ 22.54	31.35 $\pm$ 2.15
PR8		150	51.54 $\pm$ 3.31	294.33 $\pm$ 25.57	35.54 $\pm$ 2.71
PR9		200	63.57 $\pm$ 3.84	328.89 $\pm$ 36.43	46.78 $\pm$ 3.15
PR10		250	64.56 $\pm$ 4.25	556.78 $\pm$ 34.25	57.12 $\pm$ 4.12
PR11	Methocel K15M	50	42.12 $\pm$ 3.54	184.54 $\pm$ 17.63	33.13 $\pm$ 3.58
PR12		100	49.56 $\pm$ 2.67	209.72 $\pm$ 31.57	40.57 $\pm$ 3.54
PR13		150	58.64 $\pm$ 4.56	234.31 $\pm$ 24.55	51.53 $\pm$ 2.87
PR14		200	70.17 $\pm$ 3.32	258.79 $\pm$ 31.14	62.48 $\pm$ 4.83
PR15		250	68.34 $\pm$ 5.51	394.45 $\pm$ 34.73	64.59 $\pm$ 4.43
PR16	MethocelK4M	50	27.41 $\pm$ 2.47	239.47 $\pm$ 34.69	24.54 $\pm$ 1.13
PR17		100	34.26 $\pm$ 2.54	312.63 $\pm$ 14.43	31.14 $\pm$ 2.06
PR18		150	37.52 $\pm$ 3.71	353.64 $\pm$ 40.15	35.72 $\pm$ 2.47
PR19		200	43.13 $\pm$ 3.54	369.28 $\pm$ 42.28	38.18 $\pm$ 3.14
PR20		250	51.25 $\pm$ 2.35	567.61 $\pm$ 41.26	37.84 $\pm$ 2.23

\* Results are mean of three observation  $\pm$  SD

For all the batches there was a significant improvement in the particle size of microballoons. This might be attributed to presence of polymers on the surface, which increases particle-particle interaction, may cause faster squeezing out of solvents (DCM and ethanol) to the surface, which results in particle size enlargement. Results showed that microballoons obtained with EC (Ethyl Cellulose) and MethocelK15M at amount of 100 mg to 300 mg improved functionality of repaglinidemic microballoons. EC (Ethyl Cellulose) and MethocelK15M amount below 100 revealed microballoons with poor characteristics and also required more time to grow. This might be due to the inability of low amount to start formation of microballoons. Further, higher amount (>200 mg) of EC and MethocelK15M resulted in undesired PB, DG and EE. This might be due to higher viscosity in the presence of polymers. In addition to these, microballoons were carried out by utilizing different concentration of Sodium alginate (SA), Ethyl cellulose (EC), Methocel K15M (MK15M) and Methocel K4M (MK4M). However, results of SA and MK4M polymers the studied parameters were unfavourable. This might be attributed to the polymer structure, structure compatibility in term of free availability of hydrophobic

and hydrophilic group for interaction between polymer and drug, polymer molecular weight and viscosity.<sup>[6]</sup> Therefore, addition of SA and MK4M was not further considered for the preparation of Repaglinide microballoons. Furthermore, EC and MK15M were utilized for formation of repaglinidemicroballoons.

## 2. Selection of Stirring Speed

Different Repaglinidemicroballoons (PR21 to PR37) were prepared to select optimum speed of rotation (Table 3).

**Table 3: Selection of stirring speed for repaglinidemicroballoons**

Batch	Speed (rpm)	Results *		
		PB (%)	DG ( $\mu\text{m}$ )	EE (%)
PR21	200	21.66 $\pm$ 0.98	931.33 $\pm$ 123.57	16.57 $\pm$ 0.85
PR22	400	23.24 $\pm$ 1.32	721.41 $\pm$ 97.25	24.42 $\pm$ 1.26
PR23	600	32.31 $\pm$ 2.42	552.92 $\pm$ 37.86	33.54 $\pm$ 2.91
PR24	800	59.75 $\pm$ 2.65	335.47 $\pm$ 20.14	42.38 $\pm$ 3.13
PR25	1000	68.45 $\pm$ 3.56	238.34 $\pm$ 23.78	49.26 $\pm$ 2.37
PR26	1200	70.21 $\pm$ 4.24	152.41 $\pm$ 22.57	57.57 $\pm$ 3.37
PR27	1400	68.26 $\pm$ 1.24	164.56 $\pm$ 18.35	28.24 $\pm$ 2.17
PR28	1600	69.24 $\pm$ 1.42	152.56 $\pm$ 21.51	25.17 $\pm$ 2.14

\* Results are mean of three observation  $\pm$  SD

When the agitation speed was reduced to 200-600 rpm, larger irregular microballoons were obtained, where the shear energy may not be sufficient for the formation of spherical shape of microballoons. Microballoons prepared between 800 to 1200 rpm resulted in uniform size distribution. It appears to be clear that optimum shear force of the agitated liquid and collisions with equipment surfaces and other particles were squeezing and moulding the irregular microballoons into an almost perfect spherical shape. The impact of agitation speed on formation of microballoons was such that on further increasing the agitation speeds beyond 1200 rpm microballoons with undesired characteristics were produced with poor crushing strength. Therefore, it was determined that the speed between 800 to 1200 rpm was optimized by implementation of suitable design of experiment.

## 3. Selection of Temperature

Temperature is a very critical parameter in the formation microballoons. The results depicted that at low (8°C) temperature the microballoons were much bigger than at room temperature (28°C) (Table 4). Because of undesired properties of prepared microballoons of batch PR29 and PR31 they were not further considered. This might be attributed to less solubilisation of the drug and polymer would cause reduced wetting of drug particles and therefore, microballoons obtained with poor characteristics.

**Table 4: Selection of temperature for repaglinidemicroballoons**

Batch	Temperature (°C)	Results *		
		PB (%)	DG ( $\mu\text{m}$ )	EE (%)
PR29	5	27.31 $\pm$ 2.51	824.65 $\pm$ 35.31	15.31 $\pm$ 2.10

PR30	28	63.45 ± 3.24	157.24 ± 12.24	69.97 ± 3.54
PR31	40	36.46 ± 3.41	142.31 ± 21.17	57.36 ± 1.58

\* Results are mean of three observation ± SD

Results revealed that microballoons prepared at room temperature observed with good sphericity along with higher EE. While at higher temperature (40°C) microballoons were not formed with EE along with poor PB. This might be attributed to higher rate of evaporation of solvent at high temperature which leads to subsequent unavailability of bridging liquid to complete microballoons. Finally, it was decided that temperature for the preparation of microballoons was most suitable at room temperature.

#### 4. Selection of Solvent Ratio

Ratio of ethanol to dichloromethane (DCM) altered the fundamental properties of microballoons (Table 5).

**Table 5: Selection of phase ratio for Repaglinide microballoons**

Batch	Solvent ratio (Ethanol : DCM)	Results *		
		PB (%)	DG (µm)	EE (%)
PR32	1:3	43.52 ± 2.65	433.28 ± 38.68	53.65 ± 2.13
PR33	1:2	54.72 ± 5.61	322.75 ± 43.32	59.24 ± 3.41
PR34	1:1	71.43 ± 2.45	157.64 ± 23.64	71.23 ± 3.48
PR35	2:1	65.63 ± 2.34	293.68 ± 22.44	52.34 ± 2.64
PR36	3:1	54.14 ± 3.81	353.45 ± 32.76	47.42 ± 3.45

\* Results are mean of three observation ± SD

In batch PR34 (solvent ratio, 1:1), microballoons were obtained with desired characteristics like PB, DG and EE. Other batches show poor properties of microballoons and hence it was decided to use solvent ratio as 1:1 in further investigations.

#### 5. Optimization by Experimental Design

From a Quality-by-Design perspective, the aim of optimization was to establish the applicability of a Bayesian statistical methodology to identify the Design Space (DS) of agglomeration process. Following the ICH Q8 guideline, the DS is defined as the “Multidimensional combination and interaction of input variables (e.g., materials attributes) and process parameters that have been demonstrated to provide assurance of quality”<sup>[18, 19]</sup>. The majority of scientists now routinely use experimental design as part of scientific approach in order to: reduce costs, reduce waste, improve quality; reduce timelines of process investigation; obtain robust products and processes.<sup>[20]</sup> Response surface methodology (RSM) consists of a group of mathematical and statistical techniques used in the development of an adequate functional relationship between a response of interest and a number of associated control (or input) variables.<sup>[21, 22]</sup> Thus, the present investigation was persisted with Box–Behnken design for optimization of Repaglinide microballoons.

The Box–Behnken design was specifically selected since it requires fewer treatment combinations than a Central Composite Design in case involving three or four factors.<sup>[23, 24]</sup> On the basis of preliminary trials, solvent ratio of ethanol to DCM is 1:1 and

system temperature at 28 °C was fixed for all the experimental design batches. Further, EC and MK15M were optimized between 100 to 200mg along with rotational speed between 800 to 1200 rpm. There were three critical components [Amount of EC ( $X_1$ ), Amount of Methocel K15M ( $X_2$ ) and stirring speed ( $X_3$ )] significantly influenced various properties of repaglinidemicroballoons and hence, they were utilized for systemic studies (Table 6).

**Table 6: Actual levels of Box-Behnken design for Repaglinidemicroballoons**

Coded Values	Actual Values		
	$X_1^a$	$X_2^b$	$X_3^c$
-1	100	100	800
0	150	150	1000
1	200	200	1200

<sup>a</sup>  $X_1$  – Amount of EC (mg), <sup>b</sup>  $X_2$  – Amount of Methocel K15M mg), <sup>c</sup>  $X_3$  –Stirring speed (rpm) For all experimental design batches of repaglinidemicroballoons, various critical quality attributes like percent buoyancy (PB, %), mean geometric diameter (DG,  $\mu\text{m}$ ) and entrapment efficiency (EE, %) were evaluated (Table 7).

**Table 7: Results of Box–Behnken design batches of Repaglinide microballoons \***

Batch Code	$PB^a$ (%)	$DG^b$ ( $\mu\text{m}$ )	$EE^c$ (%)
B1	69.98 $\pm$ 2.75	196.15 $\pm$ 8.15	32.56 $\pm$ 2.31
B2	55.35 $\pm$ 1.86	267.33 $\pm$ 10.14	71.56 $\pm$ 1.97
B3	74.53 $\pm$ 2.35	245.19 $\pm$ 9.31	51.42 $\pm$ 1.25
B4	84.36 $\pm$ 3.04	154.87 $\pm$ 2.35	45.75 $\pm$ 2.35
B5	55.14 $\pm$ 1.17	321.41 $\pm$ 12.35	37.56 $\pm$ 2.07
B6	56.67 $\pm$ 2.35	276.2 $\pm$ 10.64	57.25 $\pm$ 2.14
B7	71.35 $\pm$ 3.21	195.32 $\pm$ 9.45	40.01 $\pm$ 2.38
B8	65.18 $\pm$ 2.68	209.45 $\pm$ 8.45	63.12 $\pm$ 3.21
B9	59.35 $\pm$ 2.14	241.75 $\pm$ 7.19	54.26 $\pm$ 2.74
B10	58.34 $\pm$ 1.87	275.31 $\pm$ 10.38	46.58 $\pm$ 3.13
B11	64.15 $\pm$ 1.75	212.45 $\pm$ 7.25	75.56 $\pm$ 2.15
B12	75.42 $\pm$ 2.31	174.32 $\pm$ 5.45	54.61 $\pm$ 2.34
B13	67.35 $\pm$ 2.23	181.55 $\pm$ 6.57	56.72 $\pm$ 3.17

\* Results are mean of three determinations  $\pm$  SD; <sup>a</sup> Percent buoyancy; <sup>b</sup> Mean geometric diameter; <sup>c</sup> Entrapment efficiency

The values of the coefficients of Equation 1 were determined by multiple linear regression analysis. <sup>[23]</sup> They are shown in Table 8. The polynomial terms could be used to draw conclusions after considering magnitude of coefficients and mathematical sign it expresses either positive or negative. Main effects represent the average result of changing one factor from -1 to +1, and the interactions explained the result when any two

or all three factors were changed simultaneously. If none of the factors had any effect, then the responses ( $Y_1$ ,  $Y_2$ , and  $Y_3$ ) would be scattered randomly around their mean value ( $b_0$ ).

**Table 8: Regression analysis of Box–Behnken design batches of Repaglinide Microballoons**

Coefficients	PB( $Y_1$ )		DG ( $Y_2$ )		EE ( $Y_3$ )	
	FM <sup>a</sup>	RM <sup>b</sup>	FM	RM	FM	RM
$b_0$	67.35	65.936	181.55	209.018	56.72	52.84308
$b_1$	-1.18	–	-6.2775	–	9.51625	9.51625
$b_2$	5.4775	5.4775	-8.49875	–	-4.4475	–
$b_3$	5.825	5.825	-40.3913	-40.3913	4.70625	–
$b_{12}$	6.115	–	-40.375	-40.375	-11.1675	-11.1675
$b_{23}$	3.07	–	-17.9225	–	-3.3175	–
$b_{13}$	-1.925	–	14.835	–	0.855	–
$b_{11}$	0.7375	–	29.48625	–	-7.3325	–
$b_{22}$	2.9675	–	4.84875	–	0.935	–
$b_{33}$	-6.0025	–	39.55875	29.25825	0.0975	–

Reduced model; <sup>a</sup>Nonsignificant ( $P>0.05$ ) coefficients for  $Y_1$ ; <sup>d</sup>Nonsignificant ( $P>0.05$ ) coefficients for  $Y_2$ ; <sup>c</sup>Nonsignificant ( $P>0.05$ ) coefficients for  $Y_3$

Using 5% significance level, a model was considered significant if the  $P$ -value (significance probability value) was less than 0.05.<sup>[26]</sup> For percent buoyancy ( $Y_1$ ), coefficient  $b_1, b_{12}, b_{23}, b_{13}, b_{11}, b_{22}$  and  $b_{33}$  were found to be insignificant ( $P>0.05$ ) and therefore, these terms were separated from their full model in order to develop reduced model (Table 7). Similarly, the coefficients  $b_1, b_2, b_{23}, b_{13}, b_{11}$  and  $b_{22}$  for mean geometric diameter ( $Y_2$ ); and  $b_2, b_3, b_{23}, b_{13}, b_{11}, b_{22}$  and  $b_{33}$  for entrapment efficiency ( $Y_3$ ); were found to be insignificant ( $P>0.05$ ) and hence, these terms were removed from their respective full model in order to develop reduced model.<sup>[27]</sup>

The removal of insignificant terms was further justified by executing analysis of variance (ANOVA) test (Table 9). In order to evaluate the fit of model, values of explained variation ( $R^2$ ), usually between 0 and 1, provide excellent guidance. Acceptable values were totally dependent on the nature of the data that were being examined. In this experiment, high value of correlation coefficients for all the selected dependent variables i.e.,  $Y_1, Y_2$ , and  $Y_3$  illustrated goodness of fit. This was an inference of validation of reduced model.<sup>[28]</sup>

**Table 9: Calculation for testing the model in portions for Repaglinide microballoons**

Model	df <sup>e</sup>	SS <sup>d</sup>	MS <sup>e</sup>	$R^2$
<b>Percent buoyancy (<math>Y_1</math>)</b>				
<b>Regression</b>				

FM <sup>a</sup>	9	901.35065	100.15007	0.92901361
RM <sup>b</sup>	2	511.46905	255.73452	0.527166319
<b>Residual</b>				Fcal = 2.4261
FM	3	68.87265	22.95755	Fcritical = 8.89
RM	10	458.7543	45.87543	df = (7, 3)
<b>Mean Geometric Diameter (Y<sub>2</sub>)</b>				
<b>Regression</b>				
FM	9	27421.55	3046.839	0.966863412
RM	3	22206.17	7402.057	0.782972857
<b>Residual</b>				Fcal = 2.7747
FM	3	939.7986	313.2662	Fcritical = 8.94
RM	9	6155.184	683.9094	df = (6, 3)
<b>Entrapment efficiency (Y<sub>3</sub>)</b>				
<b>Regression</b>				
FM	9	1787.664	198.6294	0.93457
RM	2	1223.324	611.6622	0.6395
<b>Residual</b>				Fcal = 1.9327
FM	3	125.1404	41.71348	Fcritical = 8.89
RM	10	689.4803	68.94803	df = (7, 3)
FM	3	0.1158	0.0386	
RM	6	0.2659	0.0443	

<sup>b</sup> RM, Reduced model; <sup>c</sup>df, Degree of freedom; <sup>d</sup> SS, Sum of squares; <sup>e</sup> MS, Mean of squares

The fitted model, for Y<sub>1</sub>, was evaluated using priori linearity hypothesis test [29] and the results indicated that no evidence of lack of fit was observed in the 95% confidence interval, because calculated F value (Fcal) was 2.4261 and lower than their critical/tabulated value (Ftab) which was 8.89 [17,18]. Likewise, for response Y<sub>2</sub> and Y<sub>3</sub>, calculated F values was less than their respective critical values (at  $\alpha = 0.05$ ) which suggested insignificant difference amongst full and reduced model (Table 9). The data of all the 13 batches of experimental design were used to generate interpolated values with the assistance of contour and perturbation plots. [32]

#### A. Influence of formulation composition on percent buoyancy (Y<sub>1</sub>):

The results of regression analysis for the percent buoyancy (PB) depicted positive sign for regression coefficients b<sub>2</sub> and b<sub>3</sub> whereas it offered negative sign for coefficients b<sub>1</sub>. These suggested that with increase in amount of MK15M and stirring speed with decrease in amount of EC, the PB of repaglinidemic microballoons. A highest PB of 84.36% was observed in batch B4 with levels of X<sub>1</sub>, X<sub>2</sub> and X<sub>3</sub> as 1, 1 and 0,

respectively. The results of contour plots are illustrated in Figure 1.

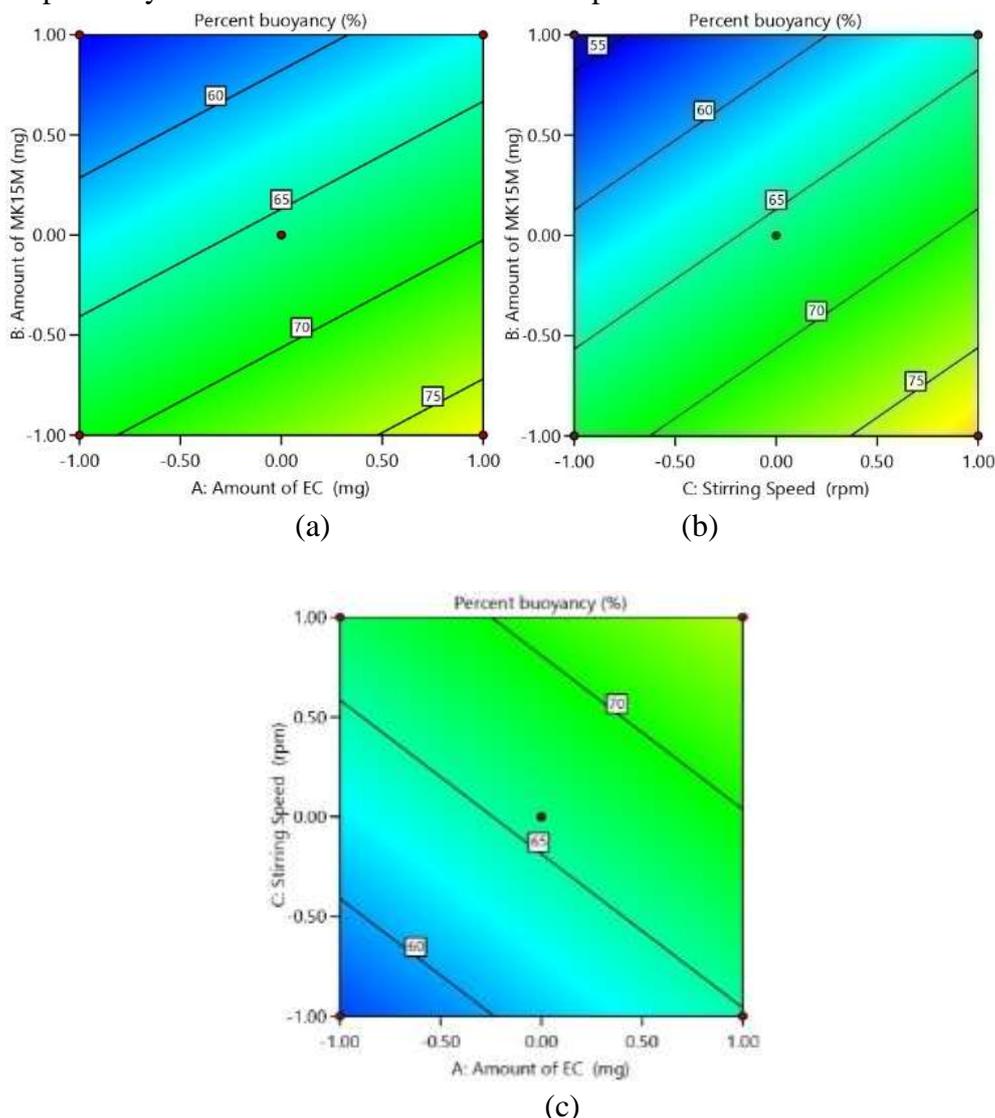


Figure 1: Influence of formulation variables  $X_1$  and  $X_2$ (a),  $X_2$  and  $X_3$  (b) and  $X_1$  and  $X_3$  (c) on PB( $Y_1$ ) of repaglidemicmicroballoons by Contour plot. From the contour plots it is clear that increased amount of MK15M and stirring speed resulted in increased the response, PB (Figure 1). Additionally, it was also found that stirring speed had little more positive effect on PB because of higher coefficient value ( $b_3 = 5.825$ ) whereas comparative low magnitude of amount of MK15M ( $b_2 = 5.477$ ) indicated the less positive effect on PB(Table 7). The value of negative coefficient ( $b_1 = -1.18$ ) of amount of EC for PBsupported significant influence.

As the number of RSM factors increased, it became difficult to visualize the response surface with graphical tools. In this case it's helpful to view a special form of response plot called "perturbation" for RSM data. Perturbation plots compare the effect of all the factors at a particular point in the RSM design space. The response was plotted by changing only one factor over its range, while holding all other factors constant. On the perturbation plot, a steep slope or curvature in an input variable revealed a relatively high sensitivity of response. Perturbation plot in Figure 2 explained the effect of variables ( $X_1$ ,  $X_2$  and  $X_3$ ) on the response, PB. The higher

negative slope of  $X_3$  indicated significant reduction in PB when deviated from -1 to 1. Similarly, positive slope of  $X_1$  and  $X_3$  revealed negative impact on PB of  $X_1$  from level -1 to 1. These outcomes validate the regression analysis (Table 7) and also confirm the findings of counter plots (Figure 1).

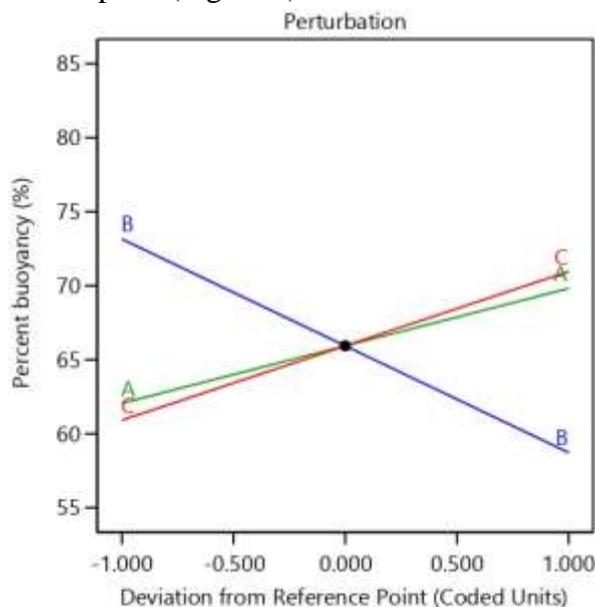
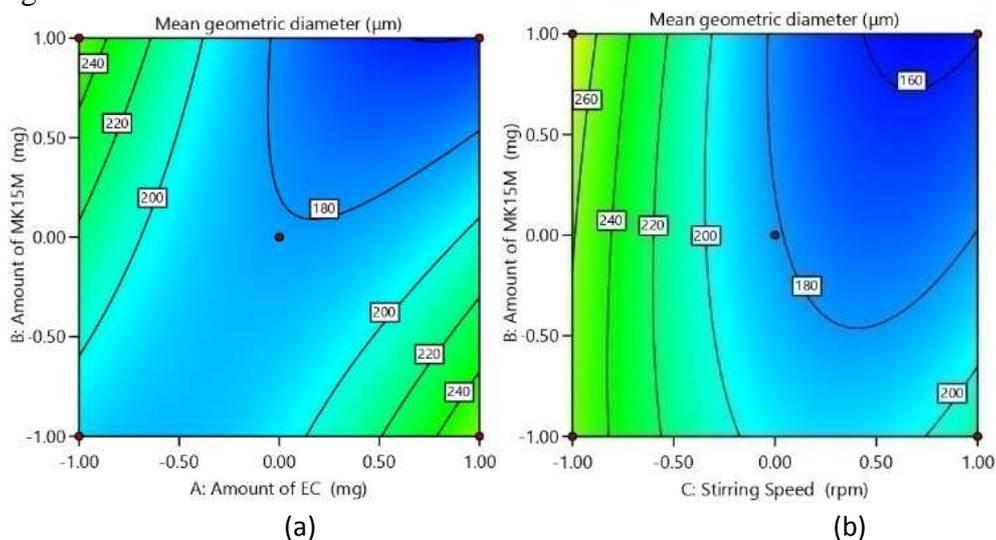
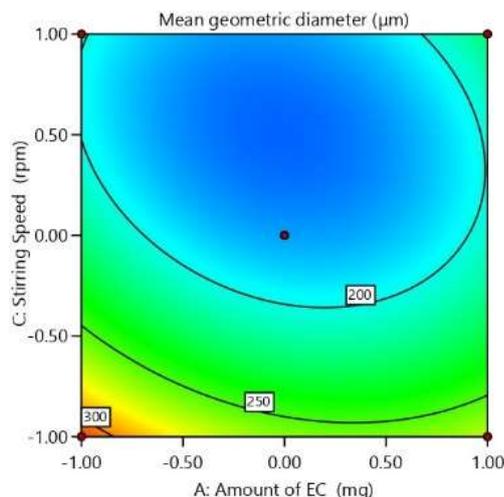


Figure 2: Perturbation plot showing effect of independent factors on PB of repaglinidemic microballoons while keeping other variables at their respective midpoint

**B. Influence of formulation composition factor on mean geometric diameter ( $Y_2$ ):** Randomly selected microballoons from each batch were subject to mean geometric diameter (DG) determination. The results of regression analysis for DG ( $Y_2$ ) depicted negative sign for regression coefficients  $b_1$ ,  $b_2$  and  $b_3$  (Table 8). These findings revealed that with decreased  $X_1$ ,  $X_2$  and  $X_3$ , decreased the DG of repaglinidemic microballoons. Greater magnitude of coefficient  $b_3$  indicated more impacts of stirring speed as compared to amount of EC and MK15M on DG. Influences of independent variables (i.e.  $X_1$ ,  $X_2$  and  $X_3$ ) on DG of microballoons are illustrated in Figure 3.





(c)

Figure 3: Influence of formulation variables  $X_1$  and  $X_2$  (a),  $X_2$  and  $X_3$  (b) and  $X_1$  and  $X_3$  (c) on DG ( $Y_2$ ) of repaglinidemicroballoons by Contour plot

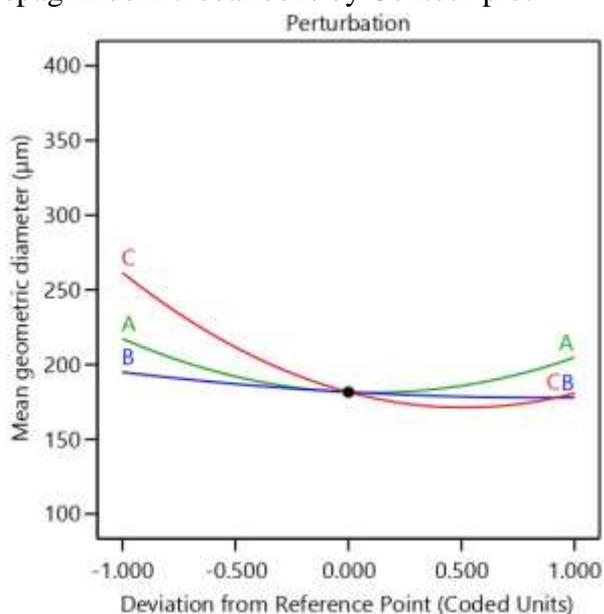


Figure 4: Perturbation plot showing effect of independent factors on DG of repaglinidemicroballoons while keeping other variables at their respective midpoint

The higher negative slope of  $X_3$  indicated significant reduction in DG when deviated from -1 to 0 whereas from 0 to -1 slope remains almost constant. Similarly, negative slope of  $X_1$  revealed negative impact on PB of  $X_1$  from level -1 to 0 whereas 0 onwards there is slight increment in DG as slope become positive to this region. Additionally, it is observed that there is very little impact of  $X_2$  on DG as slope is almost constant from level -1 to 1. These outcomes validate the regression analysis (Table 8) and also confirm the findings of counter plots (Figure 4).

### C. Influence of formulation composition factor on entrapment efficiency ( $Y_3$ ):

The results of regression analysis for  $Y_3$  depicted positive signs for regression coefficients  $b_1$  and  $b_3$ . This suggested that with increased amount of EC and stirring

speed the entrapment efficiency (EE) of repaglinidemicroballoons increased. A highest EE of 75.56% was observed in batch R11 with levels of  $X_1$ ,  $X_2$  and  $X_3$  as 0, -1 and 1, respectively. The results of contour plots are illustrated in Figure 5.

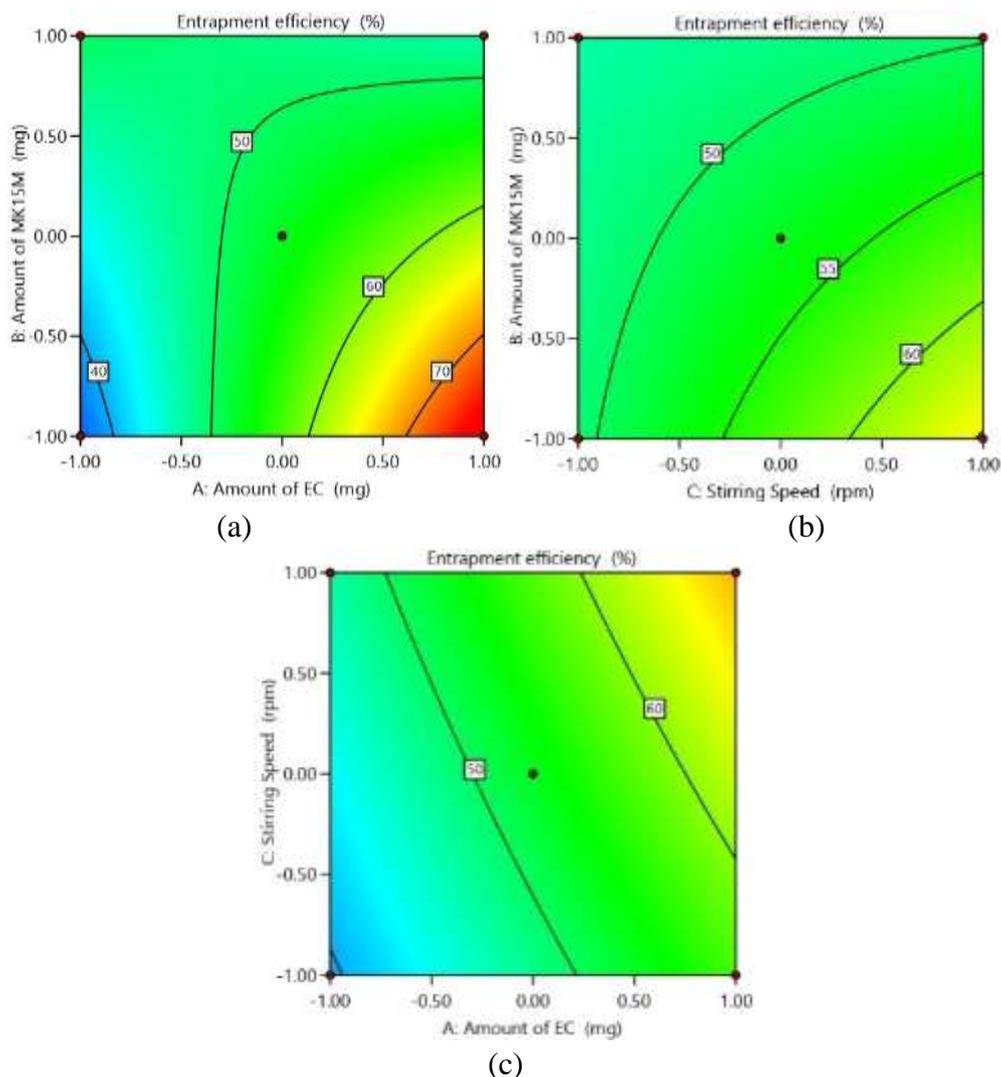


Figure 5: Influence of formulation variables  $X_1$  and  $X_2$  (a),  $X_2$  and  $X_3$  (b) and  $X_1$  and  $X_3$  (c) on EE ( $Y_3$ ) of repaglinidemicroballoons by Contour plot

Perturbation plot (Figure 6) revealed that, amount of EC and stirring speed increased the EE at broad level from -1 to 1. Additionally, as amount of MK15M increased from -1 to 1, the EE declined. This observation was might be due to increased viscosity of external phase with addition of more amount of MK15M which altered the size and distribution of microballoons. This finding also correlates the observation of counter plots (Figure 5).

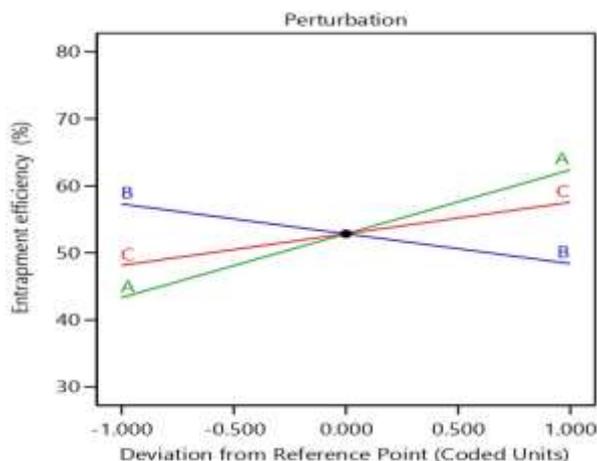


Figure 6: Perturbation plot showing the effect of each of the independent variables on EE of repaglinide microballoons

### 6. Optimization of repaglinidemicroballoons:

Optimized agglomerate in the form of check point batch was elected on the basis of the optimal solution as obtained by Design Expert<sup>®</sup>13 (Stat-Ease Inc. Minneapolis, USA) software. To optimize all the responses with different targets, a multicriteria decision approach (a graphical optimization technique by the overlay plot) was used.<sup>[31]</sup> This practice was performed within the context of the ICH Q8(R2)<sup>[34]</sup> note concerning pharmaceutical development.

The optimized formulation was obtained by applying constraints on dependent variable responses and independent variables.<sup>[35]</sup> The constraints were: maximum PB; minimal DG and maximum EE. These constraints were common for all the formulations. The recommended concentrations of the independent variables were calculated by the Design Expert<sup>®</sup> software from the overlay plot. The extensive grid and feasibility searches provided that the optimum formulations and the overlay plot is depicted in Figure 7, where one solution was found with a highest desirability. The optimum values of selected variables obtained were -0.74 ( $X_1$ ; amount of EC), 0.51 ( $X_2$ ; amount of MK15M) and 0.61 ( $X_3$ ; stirring speed).

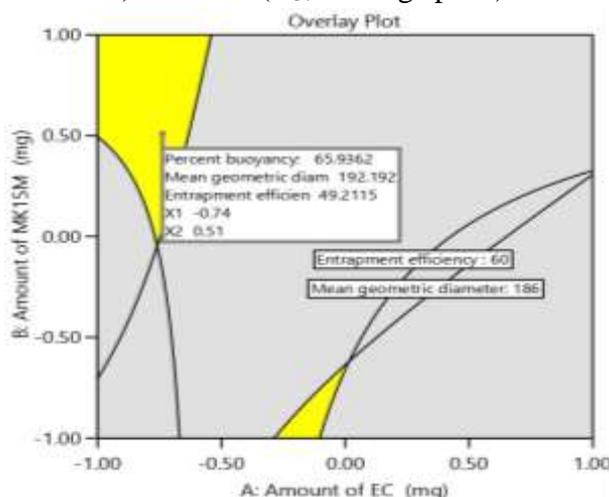


Figure 7: Overlay plot with optimized batch of repaglinidemicroballoons

Check point/optimized batch of repaglinidemicroballoons(RM) was prepared according to the levels of factors optimized. The results depicted nonsignificant ( $P>0.05$ ) difference and lower magnitude of relative error (in percent) between experimentally obtained and theoretically computed data of PB, DG and EE<sup>[36]</sup> as well as significant values of  $R^2$ <sup>[37]</sup> suggested the robustness of mathematical model and high predictive ability of RSM applied (Table 10).

**Table 10: Results of optimized batch (RM) of repaglinidemicroballoons**

<i>Response</i>	<i>Predicted Value</i>	<i>Experimental Value [Mean <math>\pm</math> SD (n=3)]</i>	<i>% Relative Error</i>
PB (%)	65.93	69.31 $\pm$ 2.12	4.88%
DG ( $\mu$ m)	192.19	181.83 $\pm$ 4.26	5.70%
EE (%)	49.21	52.47 $\pm$ 1.05	6.21%

### Stability studies

Stability is defined as the ability of particular drug or dosage form in a specific container, sealed high density polyethylene bottles, to remain within its physical, chemical, therapeutic and toxicological specification. It is considered good practice to test the stability of drug substances and drug products according to the ICH Q1A(R2) (2003); Q1B (1996), Q1C (1996), Q1D (2002), Q1E (2003) and Q5C (1995) guidelines,<sup>[39,40,41,42,43,44]</sup> or the World Health Organization (WHO) Technical Report, "Stability testing of active pharmaceutical ingredients and finished pharmaceutical products" (WHO, 2009). The ICH document Q1A(R2) states a minimum of 6 months for accelerated, stability data is needed.<sup>[38]</sup> To interpretation of stability data, simple concept of "change-over-time" and "variability" determined in literature.<sup>[45]</sup>

In the present work, accelerated stability study was carried out for selected microballoons (RM) at  $40 \pm 2^\circ\text{C}$  and  $75 \pm 5\%$  RH for six months using a programmable environmental test chamber. Various evaluation parameters were determined during the stability period and results are given in Table 12. Results of the stability study of optimized microballoons of repaglinide(RM) and exhibited no remarkable change in the all selected responses at the end of 3 and 6 months.

**Table 11: Results of evaluation parameters of RM during stability periods**

<i>Parameters</i>	<i>Storage Periods</i>		
	<i>At initial time</i>	<i>3 Months</i>	<i>6 Months</i>
PB (%)	69.31 $\pm$ 2.12	71.02 $\pm$ 2.51	70.24 $\pm$ 1.57
DG ( $\mu$ m)	181.83 $\pm$ 4.26	184.75 $\pm$ 5.04	183.02 $\pm$ 5.63
EE (%)	52.47 $\pm$ 1.05	51.98 $\pm$ 1.21	52.79 $\pm$ 0.97

\*Results are mean of three observations  $\pm$  SD

### SUMMARY AND CONCLUSION:

- Microballoons of Repaglinide can be prepared successfully by using solvent evaporation technique. The prepared microballoons exhibited a good entrapment efficiency and % Buoyancy. The concentration of Ethyl Cellulose, Methocel, surfactant, and stirring speed play a crucial role in particle size, drug entrapment efficiency, and in vitro buoyancy.

- The formulation was optimized by the Box-Behnken design, the experimental design, regression analysis, and contour plots were used in optimizing formulation variables in the preparation of Repaglinide Microballoons. The optimized formulation prepared using the predicted levels of factors provided the desired observed responses with Y1, Y2 and Y3 for percent buoyancy (PB), geometric diameter (DG) and entrapment efficiency (EE), respectively.
- The developed floating microballoons of Repaglinide increase the gastric residence time and prolong the drug release in the stomach, which, in turn, improves the local availability of the drug. From the result it can be concluded that these variables had significant effect on the responses.
- In vitro drug release studies showed that release from the microballoons get successfully retarded for over 12 h. The stability study shows no significant change in microballoons of the optimized formulation after 06 months of storage.
- The multiparticulate delivery system can prove to be a better option as compared to other oral dosage forms. Finally, it is possible to conclude that microballoons drug delivery systems can be used as gastro-retentive drug delivery systems, reducing dosing frequency and improving patient compliance.

## REFERENCES:

1. Sudheer P, Kumar H, Thomas L and Nethravathi DR: Floating microspheres - an excellent approach for gastric retention. *J Pharm Res* 2015; 14(4): 71.
2. Kumar R, Kamboj S, Chandra A, Gautam PK and Sharma VK: Microballoons: an advance avenue for gastroretentive drug delivery system- a review. *UK J Pharm Biosci* 2016; 4(4): 29.
3. Streubel A, Siepmann J, Bodmeier R. Multiple unit gastroretentive drug delivery systems: a new preparation method for low density microparticles. *J. Microencapsul.* 2003; 20: 329-347.
4. Garala KC, Shah PH. Influence of Crosslinking Agent on the Release of Drug from the Matrix Transdermal Patches of HPMC/Eudragit RL 100 Polymer Blends, *Journal of Macromolecular Science, Part A*, 2010, 47:3, 273-281.
5. Garala KC, Patel JM, Dhingani AP, Dharamsi AT. Preparation and evaluation of agglomerated crystals by crystallo-co-agglomeration: an integrated approach of principal component analysis and Box-Behnken experimental design. *Int J Pharm.* 2013 Aug 16;452(1-2):135-56. doi: 10.1016/j.ijpharm.2013.04.073. Epub 2013 May 14. PMID: 23684660.
6. Maurya DP, Sultana Y, Aqil M, Panda BP, Ali A. Formulation and optimization of alkaline extracted ispaghula husk microscopic reservoirs of isoniazid by box-behnken statistical design. *Journal of Dispersion Science and Technology*, 2011; 32:424-432.
7. Garala KC, Patel JM, Dhingani AP, Dharamsi AT. Quality by design (QbD) approach for developing agglomerates containing racecadotril and loperamide hydrochloride by crystallo-co-agglomeration. *Powder Technology*, 123, 247:128-146.

8. Chaulang G, Patil K, Ghodke D, Khan S. Preparation and characterization of solid dispersion tablet of Furosemide with crospovidone. *Research Journal of Pharmaceutical Technology*, 2008; 1(4):386-389.
9. Batt, D., &Garala, K.C. Preparation and evaluation of inclusion complexes of diacerein with  $\beta$ -cyclodextrin and hydroxypropyl  $\beta$ -cyclodextrin. *Journal of Inclusion Phenomena and Macrocyclic Chemistry*, 2012, 77, 471-481.
10. Martin A. *Physical pharmacy*. 4th ed. Lippincott Williams & Wilkins; Philadelphia, 1993.
11. Kakran M, Shegokar R, Sahoo NG, Gohla S, Li L, Müller RH. Long-term stability of quercetin nanocrystals prepared by different methods. *Journal of Pharmacy and Pharmacology*, 2012; 64(10):1394-1402.
12. Singh SP, Patra N, Chakraborty S, Pandit HK, Patro J, Devi V. Studies on Flowability, compressibility and in-vitro release of terminalia chebulana fruit powder tablets. *Iranian Journal of Pharmaceutical Research*, 2011; 10(1):3-12.
13. Zhang L, Chai G, Zeng X, He H, Xu H, Tang X. Preparation of fenofibrate immediate-release tablets involving wet grinding for improved bioavailability. *Drug Development and Industrial Pharmacy*, 2010; 36(9):1054-1063.
14. Pifferi G, Restani P. The safety of pharmaceutical excipients. *Il Farmaco*, 2003; 58:541-550.
15. Zhenhao D, Xingxing D, Xinyuan S, Yanjiang Q. Design and development of pharmaceutical excipients database. *World Science and Technology*, 2011; 13(4):611-615.
16. FDA, 2010. Inactive ingredients guidelines. <http://www.accessdata.fda.gov/scripts/cder/IIG>.
17. Kumar S, Chawla G, Bansal AK, Spherical crystallization of mebendazole to improve processability. *Pharmaceutical Development and Technology*, 2008; 13:559-568.
18. Lebrun P, Krier F, Mantanus J, Grohgan H, Yang M, Rozet E, Boulanger B, Evrard B, Rantanen J, Hubert P. Design space approach in the optimization of the spray-drying process. *European Journal of Pharmaceutics and Biopharmaceutics*, 2012; 80:226-234.
19. Mennini N, Furlanetto S, Cirri M, Mura P. Quality by design approach for developing chitosan-Ca-alginate microspheres for colon delivery of celecoxib-hydroxypropyl- $\beta$ -cyclodextrin-PVP complex. *European Journal of Pharmaceutics and Biopharmaceutics*, 2012; 80:67-75.
20. Owen MR, Armitage M, Chatfield M, Davies B, Emiabata-Smith D, Freeman S, Hayes D, Mann I, Ramsay T, Smith L, Squires B. A scientist's viewpoint on promoting effective use of experimental design: Ten things a scientist wants to know about experimental design. *Pharmaceutical Statistics*, 2003; 2:15-29.
21. Khuri AI, Mukhopadhyay S. Response surface methodology. *WIREs Computational Statistics*, 2010; 2:128-149.
22. Sajjia M, Benyounis KY, Olabi AG. The simulation and optimization of heat treatment of cobalt ferrite nanoparticles prepared by the sol-gel technique. *Powder Technology*, 2012; 222:143-151.

23. Villar AMS, Naveros BC, Campmany ACC, Trenchs MA, Rocabert CB, Bellow LH. Design and optimization of self-nanoemulsifying drug delivery systems (SNEDDS) for enhanced dissolution of gemfibrozil. *International Journal of Pharmaceutics*, 2012; 431:161–175.
24. Bezerra MA, Santelli RE, Oliveira EP, Villar LS, Escaleira LA. Response surface methodology (RSM) as a tool for optimization in analytical chemistry. *Talanta*, 2008; 76:965–977.
25. Armstrong NA. *Pharmaceutical experimental design and interpretation*. 2nd Ed., CRC Press, Taylor & Francis Group, Boca Raton, 2006.
26. Wass JA. First steps in experimental design - The screening experiment. *Journal of Validation Technology*, 2010, 49-57.
27. Singh B, Kumar R, Ahuja N. Optimizing drug delivery systems using systematic “design of experiments.” Part I: fundamental aspects. *Critical Review in Therapeutic Drug Carrier System*, 2005; 22(1):27–105.
28. Shah TJ, Amin AF, Parikh JR, Parikh RH. Process optimization and characterization of poloxamer solid dispersions of a poorly water-soluble drug. *AAPS PharmSciTech*, 2007; 8(2):E1-E7.
29. Faria AF, Marcellos LF, Vasconcelos JP, de Souza MVN, Júnior ALS, do Carmo WR, Diniza R, de Oliveira MLA. Ethambutol analysis by copper complexation in pharmaceutical formulations: spectrophotometry and crystal structure. *Journal of the Brazilian Chemical Society*, 2011; 22(5):867-874.
30. Gohel M, Patel M, Amin A, Agrawal R, Dave R, Bariya N. Formulation design and optimization of mouth dissolve tablets of nimesulide using vacuum drying technique. *AAPS PharmSciTech*, 2004; 5(3):1-6.
31. Zidan AS, Mokhtar M. Multivariate optimization of formulation variables influencing flurbiprofen proniosomes characteristics. *Journal of Pharmaceutical Sciences*, 2011; 100(6):2212-2221.
32. Mashru PP, Sutariya VB, Sankalia MG, Parikh PP. Development and evaluation of fast-dissolving film of salbutamol sulphate. *Drug Development and Industrial Pharmacy*, 2005; 31:25–34.
33. Srikanth MV, Rao NS, Sunil SA, Ram BJ, Kolapalli VRM. Statistical design and evaluation of a propranolol HCl gastric floating tablet. *ActaPharmaceuticaSinica B*, 2012; 2(1):60–69.
34. ICH, Q8(R2) Guidance for Industry. *Pharmaceutical development*, International Conference on Harmonization EMEA/CHMP/ICH/167068/042006, 2009.
35. Pattnaik S, Swain K, Bindhani A and Mallick S. Influence of chemical permeation enhancers on transdermal permeation of alfuzosin: An investigation using response surface modeling. *Drug Development and Industrial Pharmacy*, 2011; 37(4):465–474.
36. Garala K, Patel J, Patel A, Dharamsi A. Enhanced encapsulation of metoprolol tartrate with carbon nanotubes as adsorbent. *Applied Nanoscience*, 2011; 1(4):219-230.
37. Roy P, Shahiwala A. Statistical optimization of ranitidine HCl floating pulsatile delivery system for chronotherapy of nocturnal acid breakthrough. *European Journal of Pharmaceutical Sciences*, 2009; 37:363–369.

38. ICH Q1A (R2), Stability testing of new drug substances and products. 2003.
39. ICH Q1B, Stability testing: photostability testing of new drug substances and products. November 1996.
40. ICH Q1C, Stability testing for new dosage forms. November 1996.
41. ICH Q1D, Bracketing and matrixing designs for stability testing of new drug substances and products. February 2002.
42. ICH Q1E, Evaluation for stability data. February 2003.
43. ICH Q3C (R5), Impurities: Guideline for residual solvents, Geneva, March 2011.
44. ICH Q5C, Quality of biotechnological products: stability testing of biotechnological / biological products. November 1995.
45. Bar R. Statistical evaluation of stability data: Criteria for change-over-time and data variability. *PDA Journal of Pharmaceutical Science and Technology*, 2003; 57:369-377.