

# CO<sub>2</sub> CAPTURE FROM COAL FIRED ELECTRIC POWER GENERATION IN CHINA

## Zhang Yongmei<sup>[a]\*</sup>

Keywords: CO2; capture; coal fired electric power generation; china

Nowadays environmental issues due to emission of greenhouse gases such as CO<sub>2</sub> are discussed. CO<sub>2</sub> capture has also been introduced. Large amount of CO<sub>2</sub> is discharged from Chinese coal fired electric power generation, so CO<sub>2</sub> capture is mainly studied in order to protect the environment. Four types of CO<sub>2</sub> capture technologies such as decarbonation before burning, CO<sub>2</sub> capture after burning, oxygen-rich technologies and chemical looping combustion have been discussed. CO<sub>2</sub> capture is improved by using the above methods. The complete development of CO<sub>2</sub> capture technologies has resulted in good economic and social benefits around the world.

\* Corresponding Author Fax: 86-24-56860869

E-Mail: zh6688551@163.com

[a] Liaoning Shihua University, Fushun, Liaoning, P.R. China.

#### Introduction

More and more people around the world are paying attention to the effect of greenhouse gases (CO<sub>2</sub>, CH<sub>4</sub>, NO<sub>x</sub>, hydrofluorocarbon, perfluocarbon and SF<sub>6</sub>) on the environment. The effects of global warming mainly come from CO<sub>2</sub>. Chinese Government promises to reduce 40-45% of the amount of CO<sub>2</sub> emissions by 2020 based on 1.8 billion tons of CO<sub>2</sub> discharge in 2005. Ratio of coal used as a fuel to other energies is 7:3 in China. Furthermore, coal used to generate electricity is above half of total coal consumption. Coal fired electric power generation is one of the main source of CO<sub>2</sub> discharge places in China, so it is very important to control the amount of CO<sub>2</sub> discharge from the coal fired electric power generation.

In the present paper, four types of CO<sub>2</sub> capture technologies such as decarbonation before burning, CO<sub>2</sub> capture after burning, oxygen-rich technologies and chemical looping combustion are discussed. Furthermore, the optimal CO<sub>2</sub> capture method has been also pointed out.

#### **Discussion**

### **Decarbonation before burning**

Wang Xiaoliang<sup>3</sup> introduced the reaction principle of decarbonation before burning. It meant using the appropriate method to get rid of carbon from a feedstock before burning, and then carbon with energy was separated with other materials in order to take off carbon from the feedstock. An integrated gasification combined cycle (IGCC) is a technology that uses a gasifier to turn coal and other carbon based fuels into gas—synthesis gas (syngas) <sup>4</sup>. Integrated gasification combined cycle systems were advanced type systems. They consisted of air separation unit, gasifier, syngas purification units, water gas shift reactors and CO<sub>2</sub> separation unit. Combined cycle systems encompassed combustor, compressor, heat recovery steam generator and steam turbine, etc. Figure 1 showed an image for an integrated combined gasification.

Coal was sent to a rod mill to produce slurry, which was pumped into a gasifier to burn and generate syngas whose content was CO and  $H_2$ . Syngas was transported to a reforming unit where it was transferred into  $CO_2$  and  $H_2$ . Furthermore,  $CO_2$  and  $H_2$  were separated.  $H_2$  was burnt at a combustor, so this is the best way to reach zero discharge. This system not only got rid of  $H_2S$  but also decreased investment expenditures and operating fees.

#### CO<sub>2</sub> capture after burning

CO<sub>2</sub> capture after burning meant that CO<sub>2</sub> was separated from flue gas. Three types of methods for separating CO2 were the chemical absorption method, the membrane separation method and the cryodistillation.5 The solvent absorption method is one of the most popular chemical methods in the chemical plant. Figure 2 presented a diagram for CO<sub>2</sub> capture after burning.<sup>6</sup> These have the advantages of the solvent absorption method, adding CO2 to capture unit without changing original units and widely used at coal fired electric power generation. On the other hand, this method had a lot of disadvantages such as consuming a lot of solvent, pretreating flue gas (getting rid of S, NOx and particles) before CO2 capture, getting more volume of flue gas due to the existence of nitrogen, investing more money because of CO<sub>2</sub> capture and requiring high energy for separating CO<sub>2</sub> at high - pressure and high - temperature. Although CO2 capture after burning was very good, it was very difficult to use this method in chemical plants due to poor profits.

#### Oxygen-rich technologies

Wang Xiaoliang<sup>3</sup> introduced the principle of oxygen-rich technologies and compared between burning coal in an oxygen-rich environment and conventional coal combustion. The content of  $CO_2$  in the flue gas was between 8 % and 16 % during the conventional coal combustion. It was very difficult to separate  $CO_2$  due to low content of  $CO_2$  in the flue gas and more investments were required. How to increase the content of  $CO_2$  in the flue gas was one of the important factors to decrease the energy of  $CO_2$  capture. Oxygen-rich technologies also called combustion in an  $O_2/CO_2$  mixture or an air separation/ flue gas recycle technology. High purity oxygen obtained by using the air separation and partial flue gas instead of air reacted with coal, so the content of  $CO_2$  in the flue gas was improved.

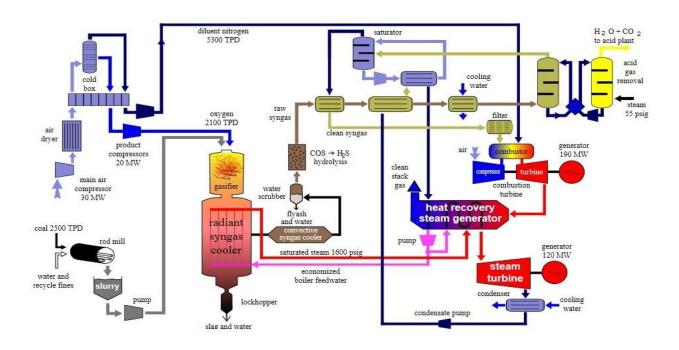


Figure 1. An integrated gasification combined cycle

Reusing the flue gas adjusted the boiler's temperature. At the same time the boiler thermal efficiency was increased due to reusing the flue gas instead of nitrogen in the air. Oxygen-rich technologies not only got high purity oxygen but also controlled coal pollution.

They are new type of coal burning technologies. Oxygen-rich technologies consisted of four types of combustion technologies such as aerobic combustion, oxygen combustion, oxy-fuel combustion and air-oxygen combustion. These advantages for oxygen-rich technologies were listed as follows. The content of

 ${\rm CO_2}$  in the flue gas reached 95 % and  ${\rm CO_2}$  was directly liquefied to recovery without the flue gas separation. High purity  ${\rm SO_2}$  existed in the boiler due to reusing the flue gas, so it improved desulfuration efficiency. The amount of the flue gas discharge was decreased about 80 % due to reusing the flue gas, so the heat loss was decreased and the boiler efficiency was improved. The radiant efficiency for burning coal in an oxygen-rich environment is better than that of the conventional coal combustion because high purity  ${\rm CO_2}$  and  ${\rm H_2O}$  in the boiler made the flue gas have high specific heat and radiation coefficient. Figure 3 showed an image for oxygen-rich technologies.

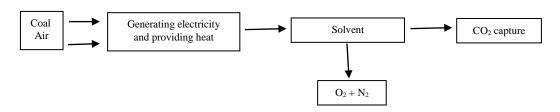
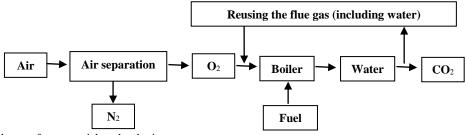


Figure 2. General scheme of CO<sub>2</sub> capture after burning



DOI: 10.17628/ECB.2013.2.290

Figure 3. General scheme of oxygen-rich technologies

#### Chemical looping combustion

Lyngfelt<sup>7</sup> discussed the principle of the chemical looping combustion. The chemical looping combustion was a new type of the burning method. It broke the conventional combustion method. The chemical looping combustion did not directly burn fuel with O<sub>2</sub> in the air. Oxygen carriers (metallic oxides) were used to move between the air reactor and the fuel reactor so the chemical energy from fuel was released. Figure 4 presented a diagram for chemical looping combustion. The fuel was pumped into the fuel reactor and reacted with metallic oxides (MeO). Eqn. (1) was written as follows. CO<sub>2</sub> and H<sub>2</sub>O were discharged from the top of the fuel reactor. High purity CO<sub>2</sub> was almost obtained when H<sub>2</sub>O was condensed. The reduced metal oxide (Me) was transported from the fuel reactor to the air reactor. Me reacted with O<sub>2</sub> in the air reactor, so Me was reused several times. Eqn. (2) was written as follows.

$$(2n+m)MeO+C_nH_{2m} \rightarrow (2n+m)Me+mH_2O+nCO_2$$
 (1)

$$Me + \frac{1}{2}O_2 \rightarrow MeO$$
 (2)

These are having the advantages of the chemical looping combustion, such as getting high purity  $CO_2$ , increasing combustion efficiency due to two step chemical reactions and decreasing the heat loss and effectively controlling the produced  $NO_x$  and its discharge.

#### Conclusion

Based on the above discussion and review, four types of  $CO_2$  capture technologies such as decarbonation before burning,  $CO_2$  capture after burning, oxygen-rich technologies and chemical looping combustion have been introduced. The decarbonation before burning is one of best ways because  $H_2$  is produced to burn at a combustor and this way, probably, reaches zero discharge. Furthermore, this system not only gets rid of  $H_2S$  but also decreases investment expenditures and operating fees.  $CO_2$  capture after burning is one of the worst methods because chemical plants obtain less income.

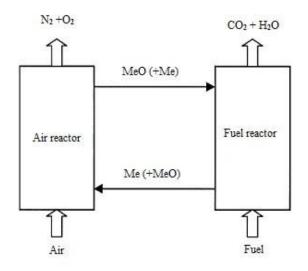


Figure 4. Diagram of chemical looping combustion

#### References

DOI: 10.17628/ECB.2013.2.290

<sup>1</sup>Yang, H. Q., Xu, Z. H. and Fan, M. H., *J. Environ. Sci.*, **2008**, 20, 14.

<sup>2</sup>Huang, B., Liu, L. B. and Xu, S. S., *Electrical Equip.*, **2008**, *19*(5), 3.

<sup>3</sup>Wang, X. L., Wu, J. H. and Zhao, Q., *Dongfang Electrical Rev.*, 2011, 25(98), 1.

4http://en.wikipedia.org/wiki/integrated\_gasification\_combined\_cycle.

<sup>5</sup>Olajire, A. A., *Energy*, **2010**, *35*(6), 2610.

<sup>6</sup>Zhang, H., Re, H. W., Lu, J. G. and Ji, Y., J. Nanjing Univ. Inform. Sci. Technol., 2009, 1(2), 129.

<sup>7</sup>Lyngfelt, A., Leckner, B. and Mattisson, T., *Chem. Eng. Sci.*, **2001**, *56*, 3101.

Received: 10.12.2012. Accepted: 31.01.2013.