



Parabens as endocrine disrupting compounds (EDCs): Analytical method for determination in environmental matrices

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Abstracts

Endocrine disrupting compounds (EDCs) discharged into the wastewater are persistent due to their continuous introduction into the environment. Recently, environmental pollution and the impacts of parabens have generated increased concern. Due to their antimicrobial and antifungal properties, parabens are a class of EDCs commonly used as preservatives in cosmetic, personal care, food, and pharmaceutical applications. The widespread uses of parabens in a variety of applications have resulted in environmental contamination. As a result, the determination of parabens using an optimized solid phase extraction (SPE) clean up together with high-performance liquid chromatography (HPLC) instruments to analyze the concentrations of parabens in environmental matrices samples was successfully documented in this paper.

Keywords: endocrine disrupting compounds (EDCs); parabens; environmental matrices; solid phase extraction (SPE); high performance liquid chromatography (HPLC)

1. Introduction

There is increasing evidence that parabens are identified as endocrine-disrupting compounds by their capacity to modulate the endocrine system (EDCs) (Wei et al., 2021). EDCs are a diverse group of chemicals that cause abnormal development by altering organism's hormonal and homeostatic systems introduced into the environment due to anthropogenic activities (Tiwari et al., 2016). EDCs can mimic or antagonize the effects of natural hormones and bind to cell receptors to block their natural functions. There are two potential mechanisms through which EDCs can interfere with hormone action. First, a chemical interacts directly with a hormone receptor; second, a hormone-receptor protein complex is directly affected (Kasonga et al., 2021). Furthermore, EDCs can potentially interfere with endogenous hormones and affect the reproductive capacity of aquatic organisms and humans; finally, can cause serious harm to the

reproductive, nervous, and immune systems of aquatic organisms (Wang et al., 2016; Deng et al., 2019).

Parabens are endocrine-disrupting compounds (EDCs) to be preservatives in personal care products, cosmetics, pharmaceuticals, and food (Bledzka et al., 2014). Parabens effectively prevent fungi, bacteria, and yeast from spoiling products and, at the same time, do not alter cosmetics or foods' fragrance, flavor, or other properties (Barabasz et al., 2019).

Methylparaben (MeP), ethylparaben (EtP), and propylparaben (PrP) are the three most frequently utilized parabens, and their usage has grown substantially throughout the years (Vale et al., 2022). Additionally, combining two or more parabens can increase their antibacterial action without exceeding their respective legal maximum concentrations (Ma et al., 2018; Alampanos & Samanidou, 2021; Wei et al., 2021). Most of the time, a mix of parabens with different solubility properties is used so that their synergistic effects can make the system more effective against contamination (Wei et al., 2021).

2. Physicochemical characteristics of parabens

Parabens are the 4-hydroxybenzoic acid alkyl esters. Based on the position of the ester groups at the C-4 position, parabens are typically divided into seven distinct types: methyl-, ethyl-, propyl-, isopropyl-, butyl-, and isobutyl-parabens. Methylparaben (MeP) and ethylparaben (EtP) are short-chain parabens, while propylparaben (PrP), isopropylparaben (i-PrP), butylparaben (BuP), isobutylparaben (i-BuP), and benzylparaben (BzP) (Haman et al., 2015). With the alkyl chain increase, parabens' water solubility diminishes while their stability, antibacterial activity, and cell penetrability rise. Because most microbial growth happens in water, parabens with a greater alkyl chain length are less effective (Alampanos & Samanidou, 2021; Wei et al., 2021).

These compounds possess unique characteristics, including low volatility, very stable that is resistant to hydrolyze, and good water solubility (Jalilian et al., 2019). Physically, parabens are white crystals or crystalline powders at room temperature; chemically, they are readily soluble in alcohol, ether, and acetone (Fransway et al., 2019). Most parabens can retain in acidic conditions. On the other hand, parabens break down into p-hydroxybenzoic acid and the corresponding alcohol in alkaline solutions (Lin et al., 2011; Bledzka et al., 2014; Li et al., 2016).

The ratio of a compound's concentration in n-octanol and water under equilibrium conditions at a particular temperature is known as the water partition coefficient (K_{ow}). Simultaneously, as the length of the alkyl chain rises, the value of the octanol-water partition coefficient (K_{ow}) increases, resulting in a decrease in water solubility. Compounds that have an increased

molecular weight readily adsorbed to sediments. Table 1 shows the physicochemical characteristics of targeted parabens.

Table 1 The physicochemical characteristics of targeted parabens

Chemical	CAS Number ^a	Molecular formula ^a	Molecular weight ^a	Water solubility (mg /L) ^b	pK _a ^b	LogK _{OW} ^b
Methylparaben	99-76-3	C ₈ H ₈ O ₃	152.14	2 500	8.31	2.00
Ethylparaben	120-47-8	C ₉ H ₁₀ O ₃	166.17	885	8.50	2.49
Propylparaben	94-13-3	C ₁₀ H ₁₂ O ₃	180.20	500	8.23	2.98

pK_a: Acid dissociation constant; K_{OW}: Octanol-water partition coefficient.

Source: ^aWei et al., (2021) , ^bLu et al., (2021)

Based on the information in Table 1, all the chemical structure of targeted parabens is shown in Figure 1.

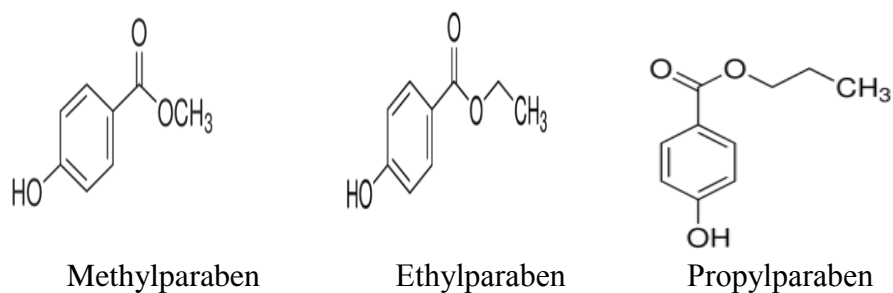


Figure 1 Chemical structure of methylparaben, ethylparaben and propylparaben

3. Sources and fate of parabens in the ecosystem

EDCs are introduced directly and inadvertently into the aquatic environment through point-source pollution from sources like industrial discharges, urban wastewater effluents, and a variety of nonpoint or diffuse sources, such as surface runoff (Noutsopoulos et al., 2019;

Kasonga et al., 2021). The primary sources include wastewater from industrial, hospital, domestic, and agricultural activities (K'oreje et al., 2020). Consequently, this could also affect drinking water sources, potentially hazardous to human health. Additionally, numerous studies have shown that personal care product residues in aquatic products impact human health (Lulijwa et al., 2020; Chen et al., 2021).

It has been demonstrated that these endocrine-mediated mechanisms have endocrine effects on the organism by interfering with multiple biological systems via distinct pathways and mechanisms (Ismail et al., 2020). Fan et al. (2021) reported that EDCs had attracted global concern in recent decades due to their significant ecological and human health risks, like deteriorating quality and quantity of human sperm, adult obesity, neurotoxicity, diabetes, and hormone-related cancers.

Most personal care products have one or two antimicrobial parabens, which are potential entry points for parabens into the ecosystem of rivers. The second possibility is that the parabens originated in the municipal sewage system. As ingredients of personal care products, pharmaceuticals, and cosmetics, parabens end up in the sewage system. These chemicals will accumulate and cause irreversible environmental changes due to their difficulty in biodegrading (Barabasz et al., 2019). Figure 2 illustrates the origins and pathways of paraben exposure in the ecosystem.

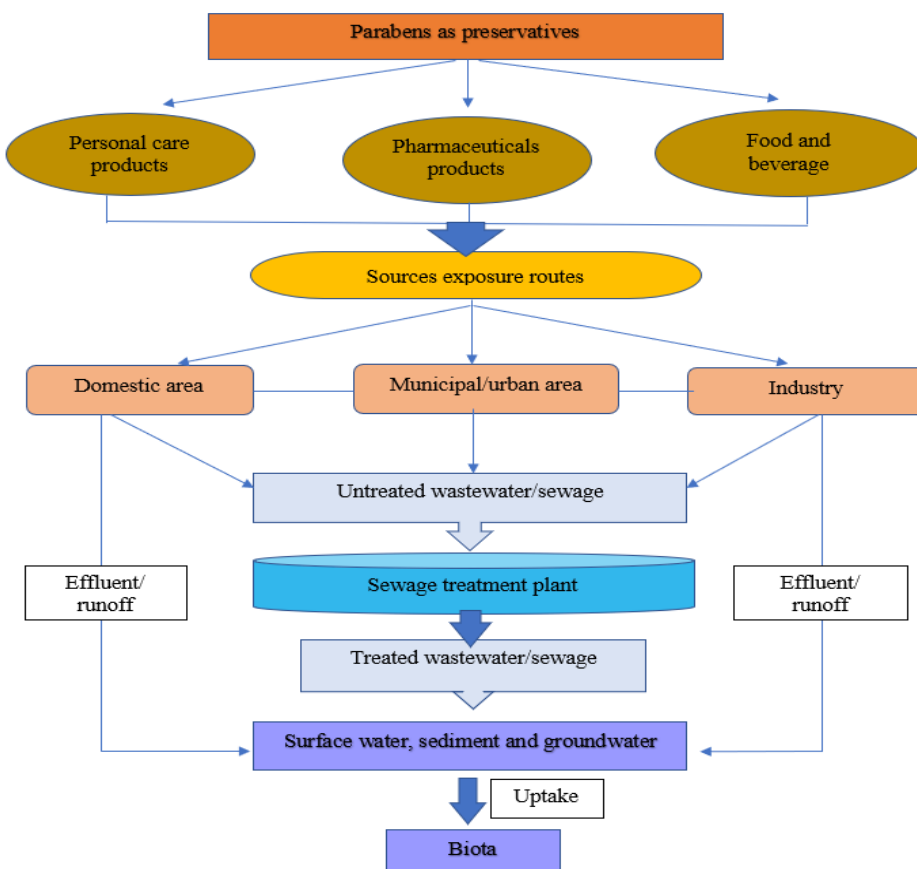


Figure 2 The origins and pathways of paraben exposure in the ecosystem.

4. Determination of parabens in environmental matrices

Prior to any analytical determination, sample preparation is required. This is because analytical instruments are not sensitive enough to find trace analytes directly in complex matrices. Target analytes were isolated and/or enriched throughout this step to prepare the samples for further analysis. The clean-up process will ensure the final extracts are clean, crucial for instrument analysis and detecting analytes at low levels. The extraction procedure is optimized to get the required accuracy and sensitivity. The aim is to minimize interferences as much as possible and eliminate co-extracted compounds.

Many techniques have been developed for effectively extracting the analyte because it is identified at low levels in aquatic environments. These techniques include liquid-liquid extraction, steam distillation extraction, and solid-phase extraction. The main advantages of the SPE method are that it has a high recovery percentage, is easy to automate, has analysis time by reducing the number of steps, works with chromatographic analysis, and uses less organic solvent. Moreover, SPE is a method of sample preparation applied in a wide variety of fields because it has many advantages over more conventional methods (Ötles & Kartal, 2016).

SPE focuses on partitioning solutes between a liquid phase (the sample matrix) and a solid phase. After extraction, solid sorbent pre-concentrates and purifies analytes from the sample. (Zwir-Ferenc & Biziuk, 2006). Since SPE requires less time and solvent, it has become the preferred method for extracting the analyte of interest and clean-up interferences from water samples (Baharom et al., 2020).

SPE is one of the most prevalent techniques for extracting pollutants from aqueous samples due to its simplest use and efficiency (Márquez-Sillero et al., 2010; Marta-Sanchez et al., 2018) It has been extensively reported for paraben extraction (Renz et al., 2013; Carmona et al., 2014). In addition, these approaches provide excellent selectivity and sensitivity while reducing or eliminating the amount of organic solvent used (Marta-Sanchez et al., 2018; Omar et al., 2021). Furthermore, SPE is able to be used in conjunction with a variety of other analysis techniques when employed to sample data. The four crucial main steps of SPE are outlined in Figure 3.

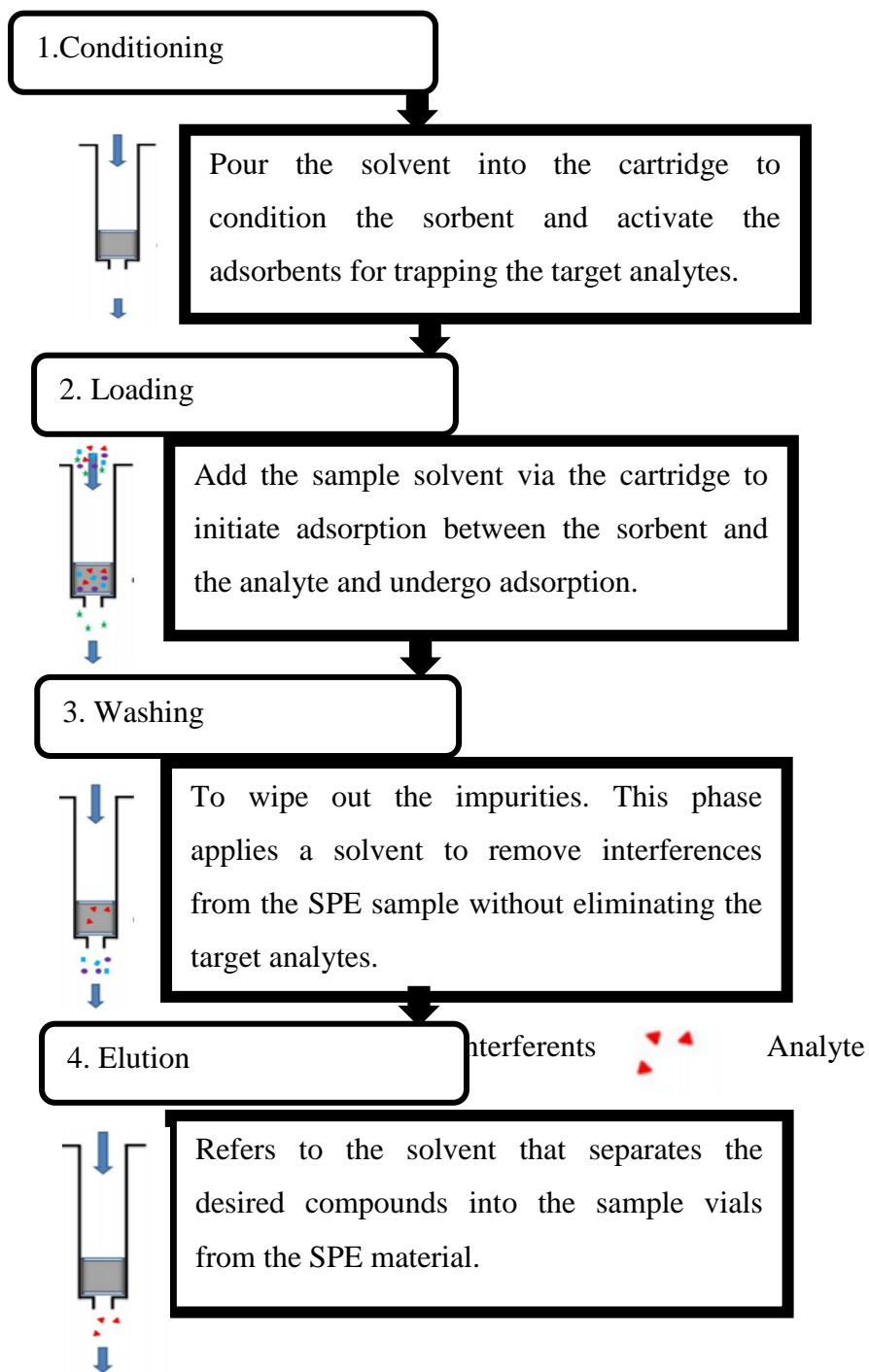


Figure 3 The four crucial main steps of SPE.

High performance liquid chromatography (HPLC) is the remarkably versatile and commonly utilized elution chromatography. This chromatographic technique was developed in the early 1970s as an alternative to GC and utilized liquid as the mobile phase (Deng et al., 2019). This technique is applied in analytical chemistry to separate, identify, and quantify each component in various organic, inorganic, and biological materials.

HPLC procedures were developed by trial and error using analyst expertise, knowledge, and judgment. Since the compounds need to be derivatized for gas chromatographic separation to proceed, HPLC techniques are more widely applied for parabens determination (Márquez-Sillero et al., 2010). In any instance, sample preparation procedures must be followed before injecting analytes into the chromatographic system.

Typical components of an HPLC instrument are a degasser, sampler, pumps, and detector. The sampler injects a sample mixture into the flow of the mobile phase. The flow of the mobile phase then moves the sample mixture to the column. Pumps ensure the mobile phase moves via the column quickly and easily. The signal produced by the detector corresponds to the sample component concentration. A digital microprocessor controls an HPLC instrument and also does data analysis. Due to its greater adaptability, HPLC has become increasingly popular for application in analyzing compounds in recent years (Locatelli et al., 2016). Figure 4 describes four basic principal parts of HPLC and an important parameter that should be considered during clean up and HPLC analysis is summarized in Table 2

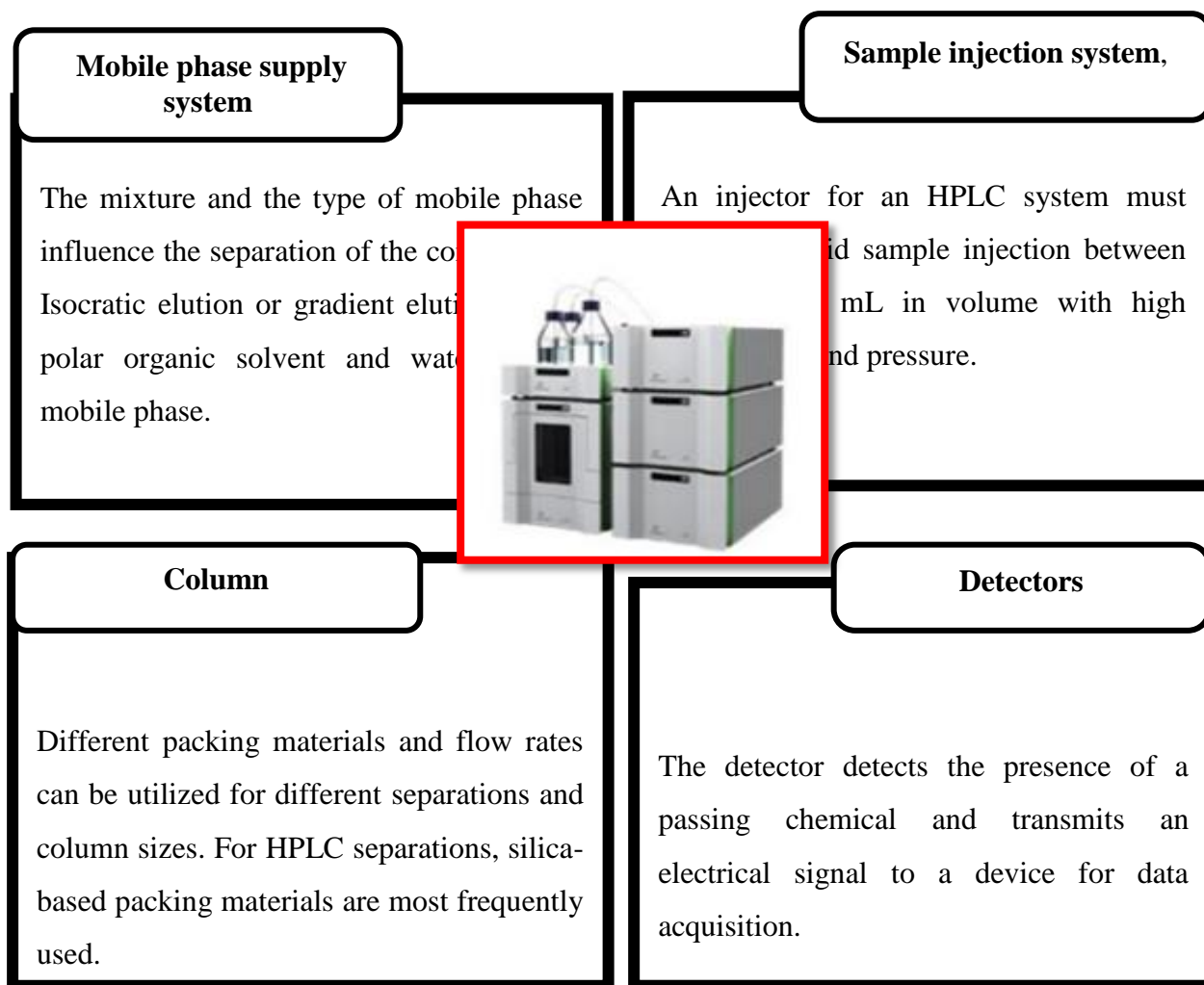


Figure 4 Four basic principle parts of HPLC

Table 2 Parameters for consideration during SPE clean up and HPLC analysis

Parameter	Function	Notes
Clean up sorbent	Reduction/elimination of matrix interferences	Choosing sorbents like polymeric C ₁₈ will help in the reduction and elimination of matrix interferences, allowing for achieving

		the lowest possible detection limit
Mobile phase	Separation of analyte	The peak separation of synthesized compounds can be improved by choosing an appropriate mobile phase composition.
Chromatographic column	Peak separation in HPLC	Selecting an appropriate chromatographic column will result in a good peak shape and the rapid elution of compounds.

Adapted from Omar et al., 2016

Figure 5 shows an overview methodology for the determination of parabens in environmental matrices. Essential steps include sample preparation and cleanup. To increase the selectivity, sensitivity, reliability, accuracy, and reproducibility of the analysis, the parabens are separated from the matrices and pre-concentrated in this step (Piao et al., 2014). The sample preparation method used largely depends on the type of matrix. Particles were removed from the water samples by filtration, and solid samples were usually homogenized, dried, and sieved before the subsequent extraction. Analytes are extracted from the sample matrix using organic solvents such as methanol, acetone, and acetonitrile for sediments and fish. In general, SPE is applied directly to liquid samples, while for solid samples, the supernatant of the analyte was extracted by ultrasonic extraction. HPLC makes the final determination as an instrument for quantifying parabens as one of the most commonly used method.

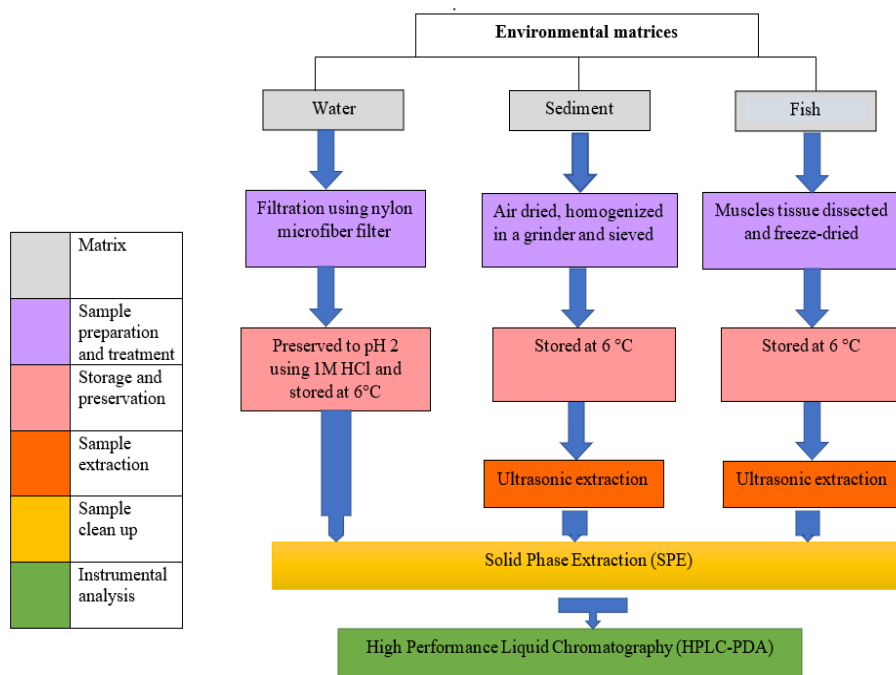


Figure 5 Overview methodology for detection of parabens in water, sediment and fish.

5. Conclusion

The quantity of EDCs that are entering the environment as a result of human activities is continuously growing. This is because of the rising demand for various products, such as pharmaceuticals and personal care items. Large quantities of parabens have entered aquatic systems via domestic and industrial wastewater due to their widespread use in consumer products. Aquatic ecosystems are particularly susceptible to anthropogenic contamination, primarily from household activities. Concerning the low concentration of parabens present in the sample matrices, it is necessary to complete the extraction and cleaning steps before the concentration is determined. Nevertheless, paraben detection helps authorities make informed decisions regarding the regulation, use, and disposal of parabens by monitoring their environmental release and impact on ecosystems.

Declaration of competing interests

The authors declare no competing financial interest.

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