

IN VITRO ANTIOXIDANT ACTIVITY OF METHANOLIC EXTRACT OF BOERHAVIA DIFFUSA LINN. (ROOTS) OBTAINED FROM FOUR DIFFERENT GEOGRAPHICAL REGIONS

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Abstract

Boerhavia diffusa Linn belonging to Nyctaginaceae family is highly reputed plant and has been widely employed in herbal medicine but no work has been carried out on comparison between the antioxidant activities obtained from four different geographical regions.

The antioxidant property of methanolic extract of Boerhavia diffusa Linn. was evaluated by estimation of Free radical scavenging activity (DPPH- radical scavenging activity), Total flavonoid contents and free reducing assay method The DPPH radical scavenging activity of **B. diffusa** was estimated by Inhibitory concentration (IC₅₀) at 517 nm and the free radical scavenging assay was analyzed by Folin - ciocalteu colorimetric method using ascorbic acid as standard, absorbance was reported at 765 nm.

Results were reported as DPPH radical scavenging activity (IC $_{50}$ value) was 350 μ g/ml and free scavenging assay was 76.85% (Bangalore) 75.22% (Jhansi),74.88% (pune),73.92% (Delhi) respectively at a concentration of 50 μ g/ml.

Thus from ongoing study it is concluded that roots of B. diffusa Linn. possess significant Antioxidant activity and can be used for further research study. But the plant which is collected from Bangalore and Pune regions has maximum in vivo antioxidant activity. These shows the geographical conditions affect the phyto constituents of plant.

Key words: in vivo Anti-oxidant activity, Boerhavia diffusa (Linn) BD, Pune (P), Bangalore (B), Jhansi (J), Delhi (D).

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1. INTRODUCTION

Boerhaavia diffusa Linn. commonly known as Punarnava. The plant has drawn lot of attention due to its uses in Indian Traditional Medicine. The various parts of the plant are used in the treatment of cancer, jaundice, dyspepsia, inflammation, enlargement of spleen, abdominal pain and as an anti-stress agent. Oxidative stress is an important risk factor in the pathogenesis of numerous chronic diseases [1,6,18]. Free radicals and other reactive oxygen species are recognized as agents involved in the pathogenesis of sicknesses such as inflammation. diabetes. diabetes. Parkinson's and Alzheimer's diseases. atherosclerosis a as well as cancers. Reactive oxygen species are also said to be responsible for the human aging.

An antioxidant can be broadly defined as inhibits substance that oxidative damage to a target molecule. The main characteristic of an antioxidant is its ability to trap free radicals. Antioxidant compounds like **Phenolics** acids. polyphenols and Flavanoids scavenge free radicals such as peroxide, hydro peroxide or lipid peroxyl and thus inhibit the mechanisms oxidative that degenerative diseases. Herbal plants considered as good antioxidant since ancient times[2,3,4].

В. diffusa Linn belonging Nyctaginaceae family is highly reputed plant and has been widely employed in herbal medicine and aromatherapy. B.diffusa Linn. is well-known for its nutrition and health-promotion values. Different parts of B. diffusa Linn. have been used for the treatment of various human ailments such as itches, cuts, ulcers, swellings, bilious fever, catarrh, eczema, .antipsychotic etc. but no work has been carried out on the antioxidant activity of the roots extracts.

Hence, the current study was designed to evaluate the antioxidant activity of

methanolic extracts of Boerhavia diffusa linn. (roots) obtained from four different geographical regions including by using DPPH scavenging assay, Total Flavonoids and Ferric Reducing Power method.

2. MATERIALS AND METHODS

Plant material

The roots of B. diffusa Linn. of all the four samples were procured from different botanical gardens and plants samples were identified and further confirmed by matching with the samples in the LWG herbarium.

Preparation of Sample:

Methanolic extract of roots of all four sample of B. diffusa Linn. were prepared through cold percolation by using 2 gm. of powdered material in 100 ml of methanol. The filtered solution were cooled and dried for the evaporation of the solvent. Then they were dissolved in methanol at concentration and used for further analysis [5].

Total Antioxidant Activity

1. DPPH Radical Scavenging Assay:

A solution of 0.135 mm DPPH in methanol was prepared and 1.0 ml of this solution was mixed with 1 ml of extract in methanol containing 0.02-0.1 mg of the extract. The reaction mixture was vortexes thoroughly and left in the dark at room temperature for 30 minutes the absorbance of mixture measured spectrophotometrically at 517 nm ascorbic acid. Ouercetin and acid were used Ascorbic references. The ability to scavenge DPPH radicals was calculated by the following equation:[19]. The result shown in table 1 and 2, Fig 1 and 2.

DPPH radical scavenging activity (%)= $[(Abs_{control} - Abs_{sample})] \times 100$ (Abs_control)

Where: Abs _{control} is the absorbance of DPPH radical + methanol; Abs _{sample} is the absorbance of DPPH radical + sample extract / reference.

2. Total Flavonoid Content:

Prepare a stock solution (1mg/ml) of extract in methanol. From the stock solution take suitable quantity of the extract into 25ml volumetric flask and add 10ml of water and 1.5ml of folin ciocalteu reagent keep the mixture for 5 min and then add 4ml of 20% Na₂CO₃ and make up to 25ml with distilled water. Keep the mixture for 30 min and record absorbance at 765nm. Total flavonoid content was calculated as Querecetin (mg/ml) using the following equation based calibration curve: y = 73.56x + 0.073, R² = 0.997, where y was the absorbance and x was Querecetin equivalent (mg/ml) [Ez Ordon L. A. A. et al., 2006]. The percentage of the total phenolic was calculated in triplicate with reference to the air dried drug.[16] The result is shown in 3 and 4.

3. Ferric Reducing Power Assay: Reducing power of extracts was determined as described by Oyaizu [1986].

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The solution of the extracts (0.2–1.0 mg) in 1 ml of distilled water was mixed with ml of 0.2 mol/L phosphate buffer (pH 6.6) and 2.5 ml of 1% (w/v) potassium

ferricyanide. The mixture was incubated at 50oC for 20 min. Following this, 2.5 ml of 10% (w/v) Trichloroacetic acid was added and the mixture was then centrifuged at 1750 rpm for 10 min. A 2.5 ml aliquot of supernatant was mixed with 2.5 ml of distilled water and 0.5 ml of 0.1% (w/v) FeCl3; the absorbance of the mixture was read at 700 nm [16,17]. The result shown in **table 5 and 6.**

3. RESULT

- **DPPH** Radical Scavenging **Activity**: Methanolic extract of B.diffusa roots extract shows maximum inhibition of radicals upto 76.85% (Bangalore) 75.22% (Jhansi),74.88% (pune),73.92% (Delhi) at concentration of 50 µg/ml.
- Concentration of 5 ug/ml & 5 ug/ml. IC₅₀ of sample was calculated as **385µg/ml**. while the standards Ascorbic acid shows maximum inhibition of 71.91% at concentration of 5 ug/ml.

Table 1: Absorbance at various concentrations (mg/ml) of in-vitro antioxidant activity:

Concentration	0.02	0.04	0.06	0.08	0.1	control
Ascorbic acid	0.5262	0.4048	0.3084	0.2333	0.4716	0.611
Quercetin	0.511	0.405	0.302	0.201	0.154	0.519
B. diffusa (B)	0.381	0.273	0.21	0.139	0.109	0.492
B. diffusa (P)	0.391	0.29	0.215	0.16	0.11	0.44
B. diffusa (J)	0.401	0.382	0.25	0.175	0.134	0.45
B. diffusa (D)	0.398	0.32	0.25	0.12	0.115	0.447

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Table 2: Percentage o	t Standards and	Sample at various	concentrations (m	g/mI)

Concentration	0.02	0.04	0.06	0.08	0.1
Ascorbic acid	13.879%	33.748%	49.525%	61.817%	71.915%
Quercetin	1.541%	21.965%	41.811%	61.272%	70.328%
B. diffusa (B)	22.561%	40.393%	54.148%	69.651%	76.201%
B. diffusa (P)	11.136%	34.091%	51.136%	63.636%	75.000%
B. diffusa (J)	10.889%	15.111%	44.444%	61.111%	70.222%
B. diffusa (D)	10.962%	28.412%	44.072%	73.154%	74.273%

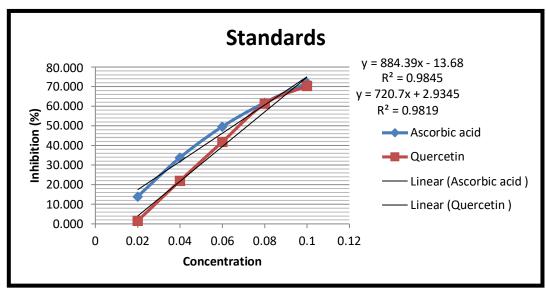


Figure 1. Graph between Absorbance and Concentration in DPPH assay

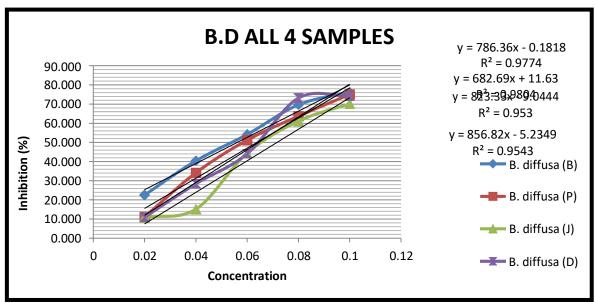


Fig 2. Graph between Absorbance and Concentration in DPPH assay

Total Flavonoid contents:

Table 3: Absorbance at various concentrations of standard and sample (mg/ml) of in-vitro antioxidant activity:

Concentration	0.004	0.008	0.012	0.016	0.020
Quericetin	0.3453	0.6692	0.9677	1.2711	1.5157
B.diffusa	0.0015	0.0018	0.0018	0.0019	0.0018
Linn. (B)					
B.diffusa	0.0015	0.0018	0.0018	0.0019	0.0018
Linn. (P)					
B.diffusa	0.0013	0.0016	0.0016	0.0016	0.0017
Linn. (J)					
B.diffusa	0.0014	0.0017	0.0016	0.0016	0.0016
Linn. (D)					

Table 4: Preparation of calibration curve for flavonoid content

S. Amoun		2%	2% Dist. aluminum Wate		Abso. At 420 nm				
no	t from stoke (ml)	chloride solution(ml	r (ml) up to	Conc. (mg/ml	B.diffus a (B)	B.diffus a (P)	B.diffus a (J)	B.diffus a (D)	
1	0.2	0.5	10	0.004	0.3453	0.3521	0.3421	0.3453	
2	0.4	0.5	10	0.008	0.6692	0.6701	0.5682	0.5986	
3	0.6	0.5	10	0.012	0.9677	0.9680	0.7456	0.8653	
4	0.8	0.5	10	0.016	1.2711	1.2692	1.3652	1.3125	
5	1	0.5	10	0.020	1.5157	1.5200	1.4510	1.4682	
6	Blank	0.5	10	-	-	-	-	-	

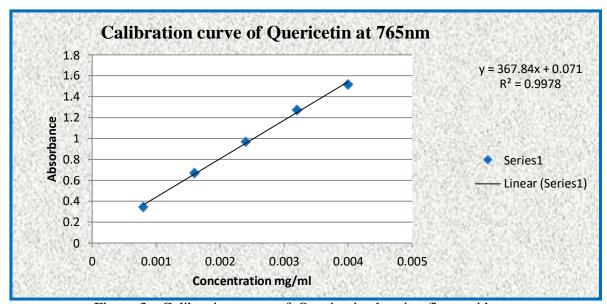


Figure 3:- Calibration curve of Quericetin showing flavonoid content

Ferric Reducing Power Assay:

Table 5: Preparation of dilutions for calibration curve (std. is Ascorbic acid).

S. No	Volume Taken From Stock (ml)	Distill Water upto 1ml	Phosphate buffer (pH 6.6) + Potassium ferricyanide (ml)	Final Volume Taken (ml)	Distill water (ml)	Ferric chloride (0.1% w/v)	Absorbance at 700 nm Ascorbic Acid
1	0.02	0.98	2.5 + 2.5	2.5	2.5	0.5	0.2363
2	0.04	0.96	2.5 + 2.5	2.5	2.5	0.5	0.2488
3	0.06	0.94	2.5 + 2.5	2.5	2.5	0.5	0.2899
4	0.08	0.92	2.5 + 2.5	2.5	2.5	0.5	0.3222
5	0.1	0.90	2.5 + 2.5	2.5	2.5	0.5	0.3539
6	Control	1.0	2.5 + 2.5	2.5	2.5	0.5	0.2363

Table 6: Absorbance at various concentrations (mg/ml) of in-vitro antioxidant activity:

Concentrat ion	Absorbance at 700 nm							
	Ascorbic acid	B.Diffusa (B)	B.Diffusa (J)	B.Diffusa (P)	B.Diffusa (D)			
1) 0 .020	0.2363	0.207	0.201	0.21	0.207			
2) 0.040	0.2488	0.22	0.21	0.215	0.223			
3) 0.060	0.2899	0.237	0.225	0.234	0.238			
4) 0.080	0.3222	0.262	0.255	0.252	0.256			
5) 0.10	0.3539	0.282	0.281	0.286	0.289			

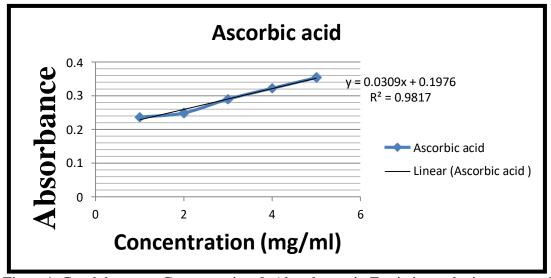
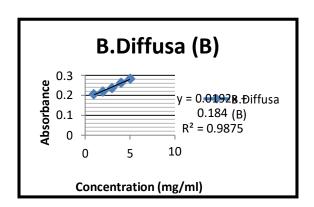
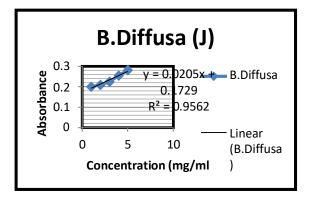
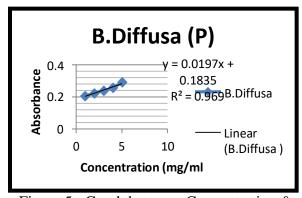


Figure 4. Graph between Concentration & Absorbance in Ferric ion reducing assay of Ascorbic acid







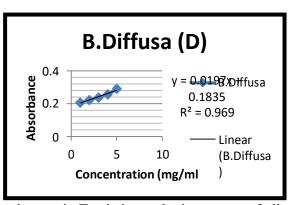


Figure 5 . Graph between Concentration & Absorbance in Ferric ion reducing assay of all B.diffusa samples

The reducing power of B. diffusa Linn extracts of four different using samples the potassium ferricyanide method is shown in Figure 4. The result indicates that the ability of reducing the extract increased with the concentration. as

shown by the increasing optical density at **700 nm**.

4. DISCUSSION CONCLUTION

AND

- In-vitro DPPH free radical scavenging activity: In-vitro DPPH free radical scavenging activity of the methanolic extract of four different samples of B.diffusa Linn. compared were Ascorbic acid and Ouericetin (standard used), methanolic extract of Boerhavia diffusa Linn. roots extract shows maximum inhibition of free radicals up to 76.201% (Bangalore) 75.0% (Pune),70.2% (Jhansi),74.273% (Delhi) respectively, at a concentration of 50 µg/ml. The two standards Ascorbic acid and Quercetin shows maximum inhibition of 71.91% & 70.32% at concentration of 5 ug/ml for both the standards. As shown in table 1,2 and figure 1 and 2.
- The DPPH anti-oxidant assay is based on the ability of DPPH a stable free radical, to decolorize in the presence of anti-oxidants. Analysis by one way ANOVA shows that value of sample is comparable to that of Standards and hence it is stated that Antioxidant activity is maximum in Bangalore and Pune region samples.
- Total flavonoid Content: In vitro **TPC** activity of the methanolic extract of roots of all the four samples of **B.**Ldiffusa Linn. collected from different geographical regions were compare with Quericetin (standard used).Results obtained present study revealed that the level flavonoind content in the methanolic extract of the roots of B.diffusa Linn. 11.322±0.0640 mg/g which was compared to Ouericetin. The result was tabulated in table 3 and 4, Fig. 3.

The reducing power of four different extracts of B.diffusa Linn. using the potassium ferricyanide method is shown in **Fig. 4.** The result indicates that the reducing ability of the extracts increased with the concentration as shown by the increasing optical density at 700 nm. The result was tabulated in **table 5 and 6**, **Fig.4and 5**.

Reducing power is associated with its antioxidant activity and may serve as a significant reflection of the anti-oxidant activity. Compounds with reducing power indicate that they are electron donors and can reduce the oxidized intermediates of lipid peroxidation processes, so that they can act as primary and secondary antioxidants

Thus, the plant have good in vivo antioxidant activity and could justify the ethnotherapeutic usage of this plant by the traditional healers. But the plant which is collected from Bangalore and Pune regions has maximum antioxidant activity. These shows the geographical conditions affect the Phyto constituents of plant.

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Competing Interest:

All of the authors have nothing to declare as far as the conflict of interest is concerned.

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